

# Scalable synthesis of lignin nanoparticles via subcritical water treatment

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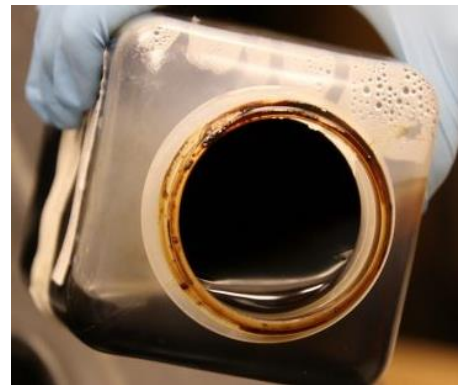
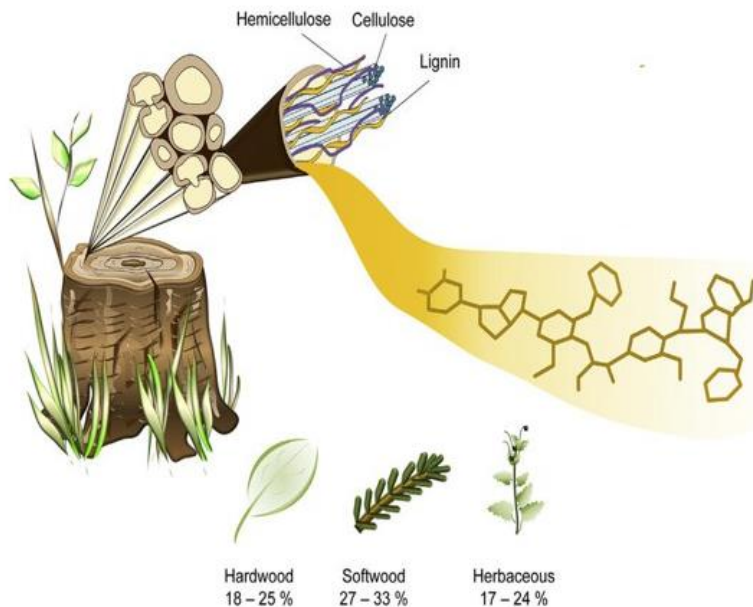
University of British Columbia, Vancouver, BC, Canada



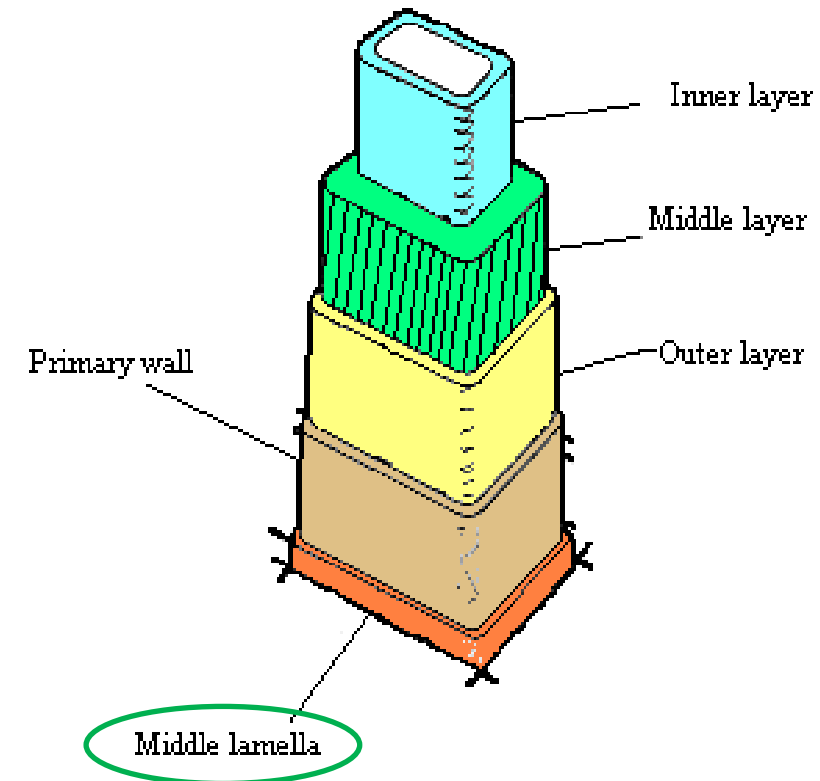
# Why Lignin?

2<sup>nd</sup> most abundant biopolymer on the planet

- 15-35% of total tree composition
- Assists in water transport, stiffness, and prevents degradation in the tree cell wall
- Kraft lignin can be extracted through precipitation and purification of black liquor
- Makes up 30-45% of the solids content of black liquor

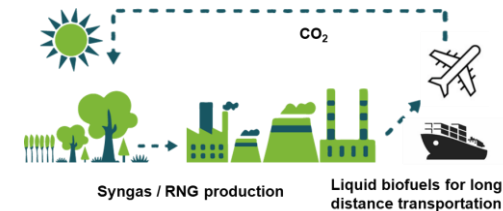
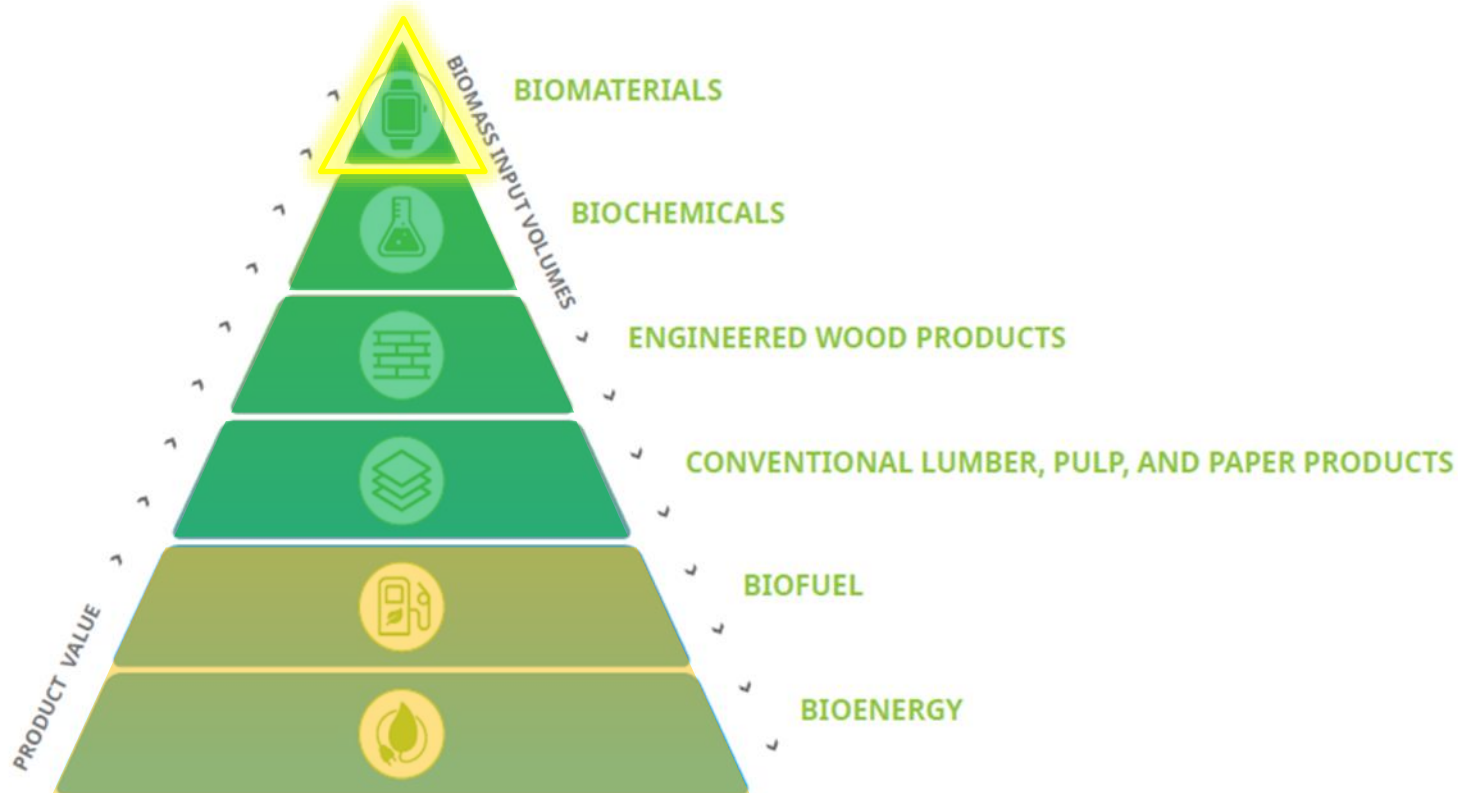


*Black liquor*



# Lignin - Uses

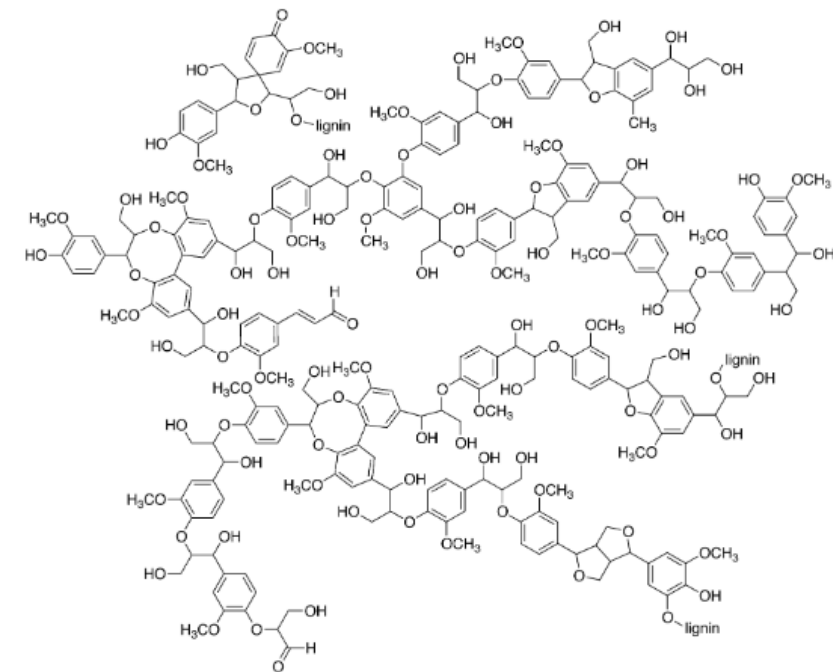
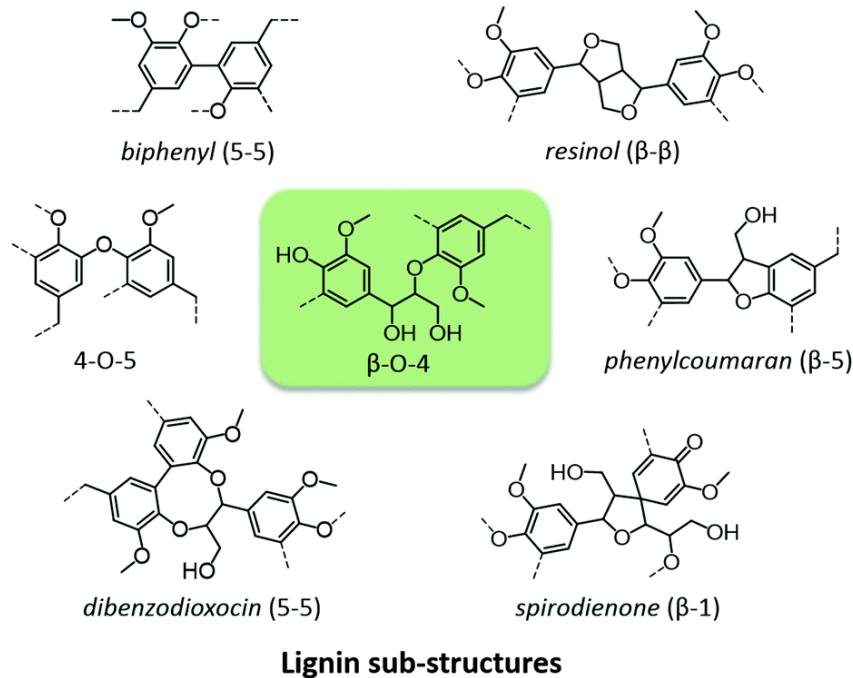
- Over 98% of the >70M tons of lignin produced annually is burned for energy
- Allows for energy self-sufficient pulp mills



Source:

# Commercializing Lignin?

Heterogenous, complex chemical structure & irregular morphology limit applications

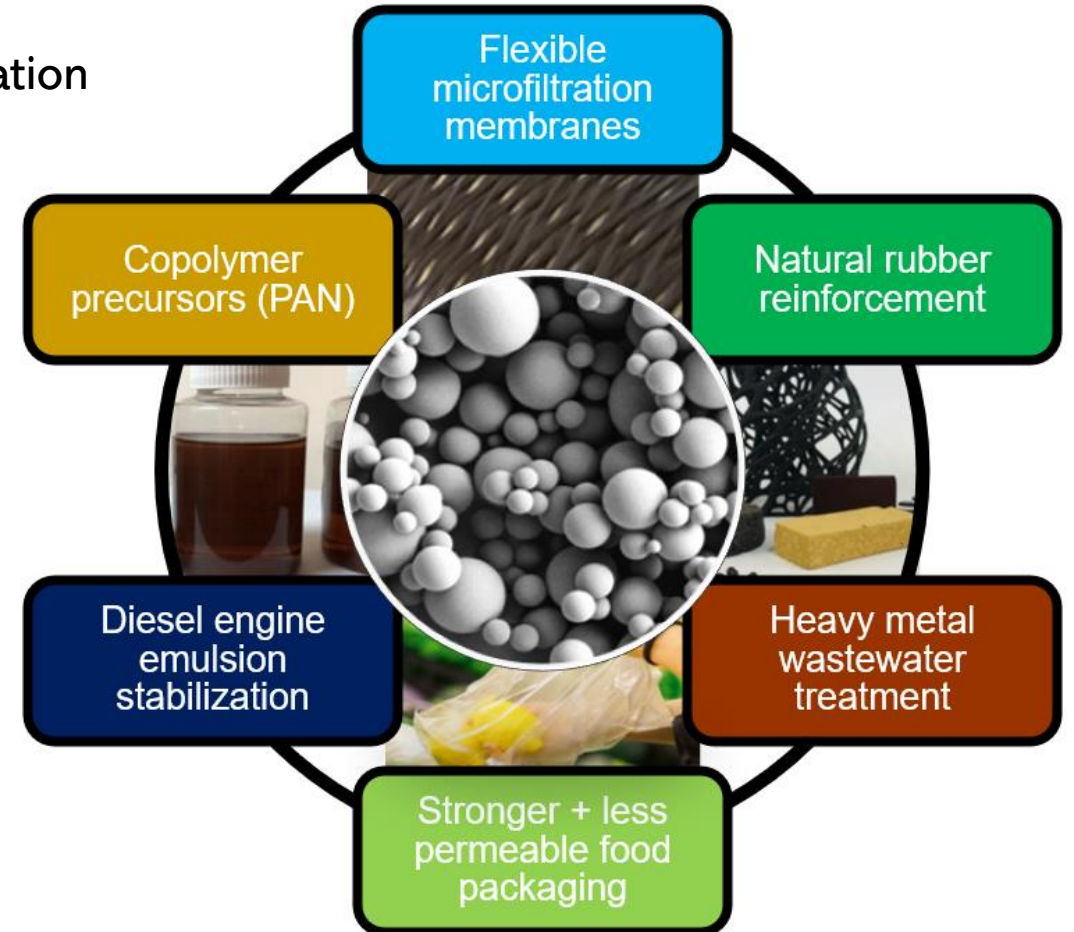
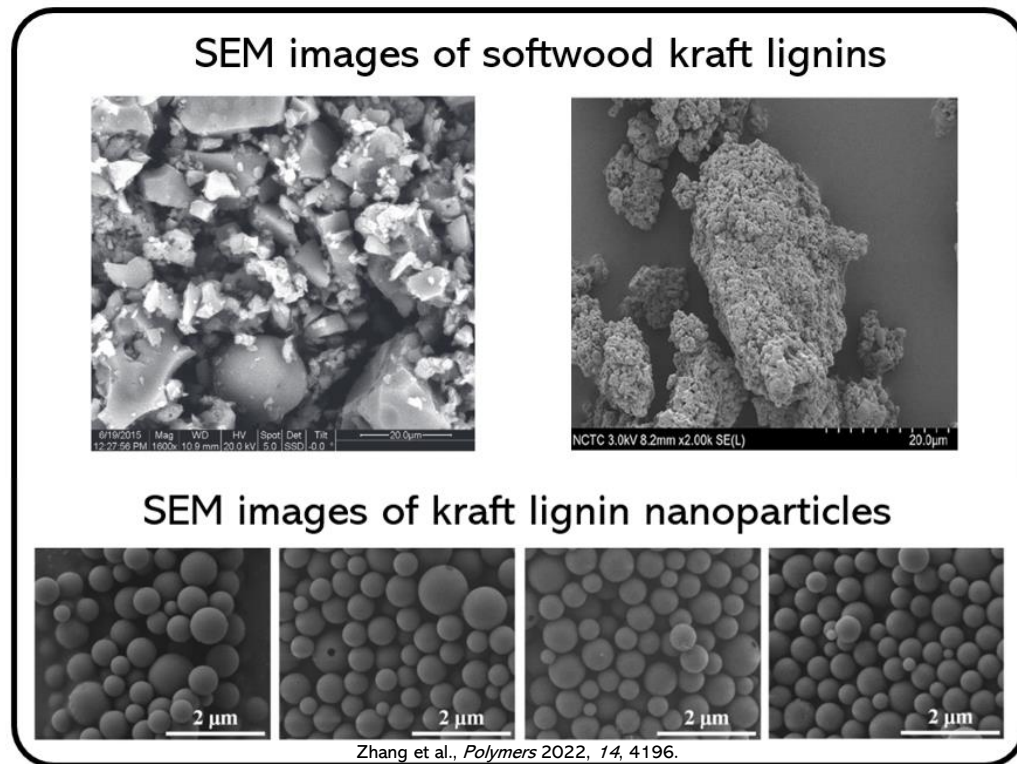


But what if we could add value to kraft lignin for  
**high performance commercial applications?**

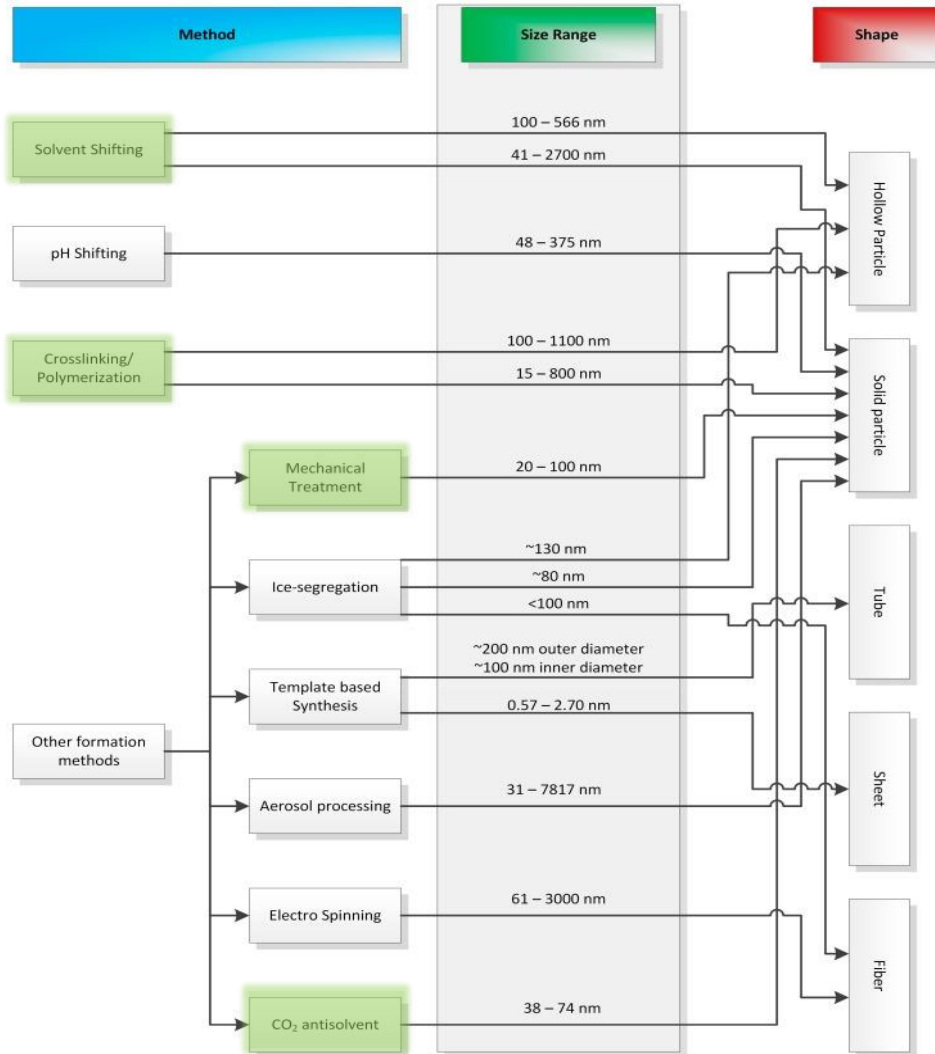


# Spherical Lignin Nanoparticles (LNPs)

- Spherical LNPs enable standardized morphology
  - Uniformity
  - Surface-area to volume ratio
  - Functionalization
  - Solubility



# Commercialization of LNPs?



Methods of LNP synthesis

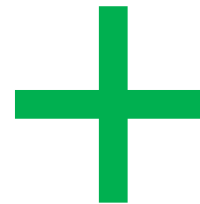
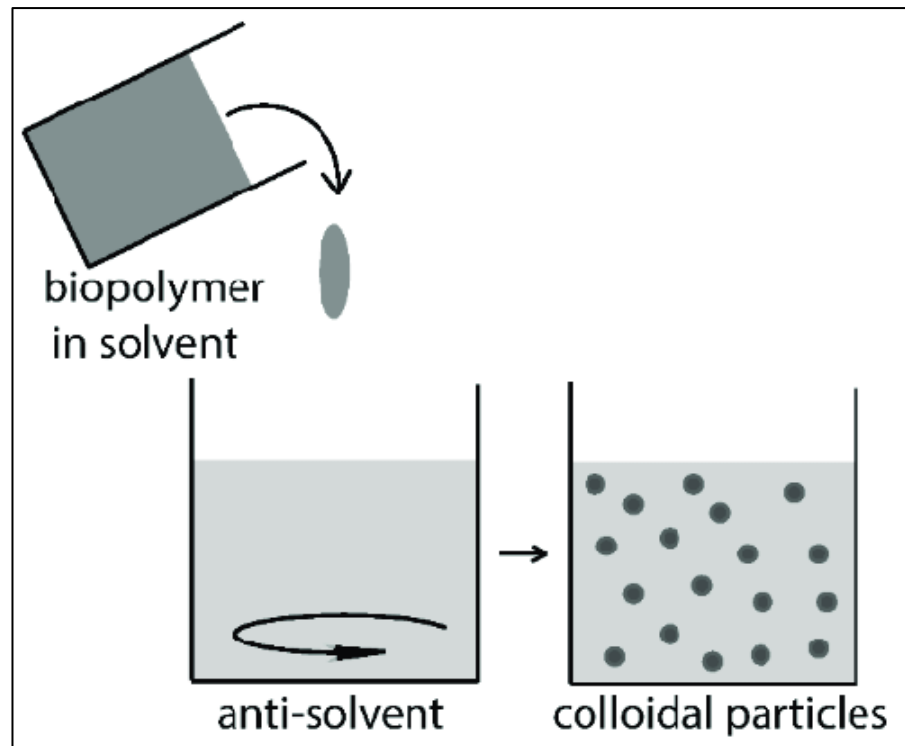
## Challenges

- Expensive
- Organic solvent-dependant
- Multi-step processes
- Specialized equipment
- Low yield
- Non-uniform product

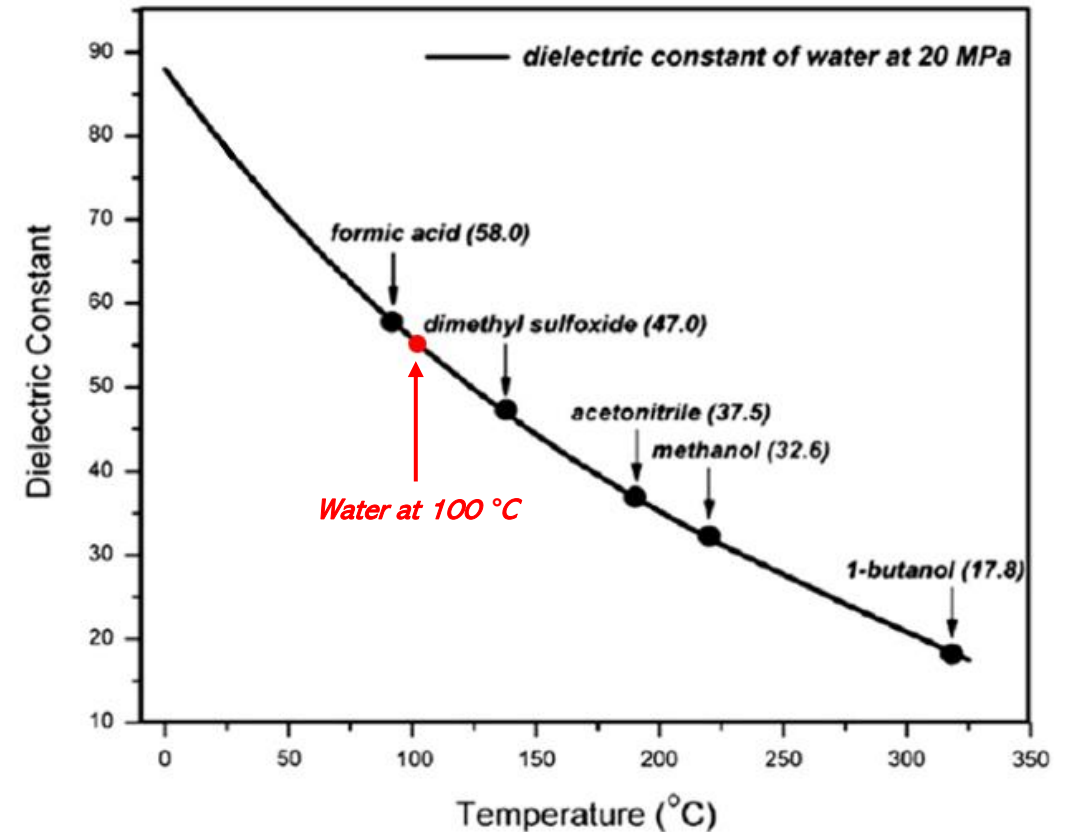
An earth-friendly, cost-effective approach is necessary to consider lignin nanoparticles for commercialization.

# LNPs from scH<sub>2</sub>O: Theory

Solvent-anti-solvent (SAS)  
precipitation process

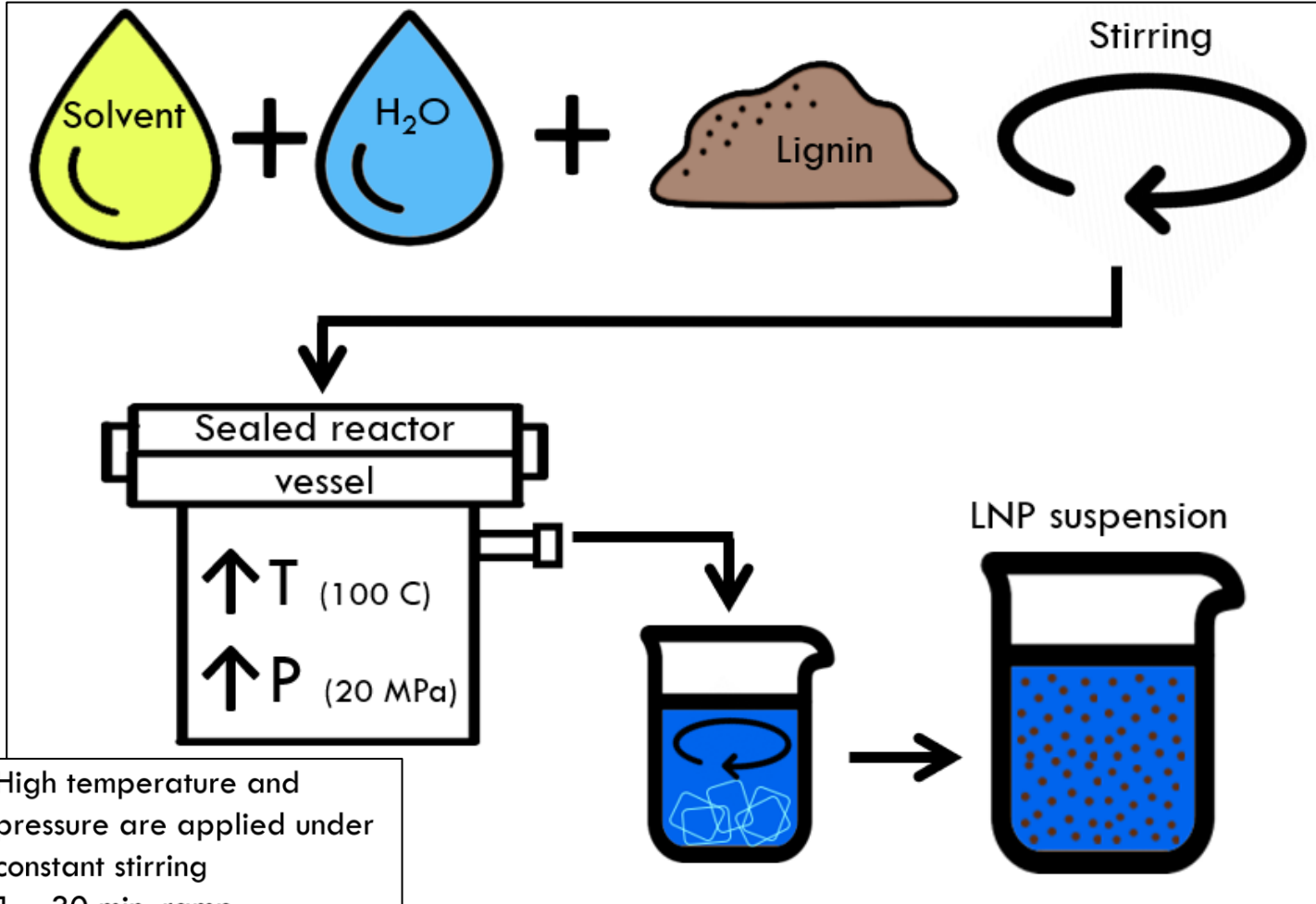


scH<sub>2</sub>O\* as a polar organic  
solvent



\*: subcritical water (scH<sub>2</sub>O): 100°C – 374 °C, P > H<sub>2</sub>O vapor pressure

# Experimental Procedure



High temperature and pressure are applied under constant stirring

1. 30 min. ramp
2. 60 min. steady
3. Rapid quenching

## Problem:

Lignin is not 100% soluble in  $scH_2O$ , creating heterogenous nucleation sites upon SAS mixing

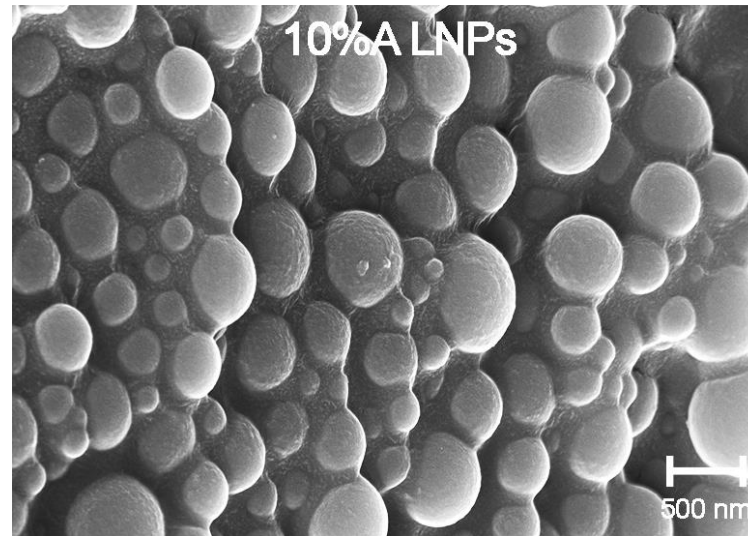
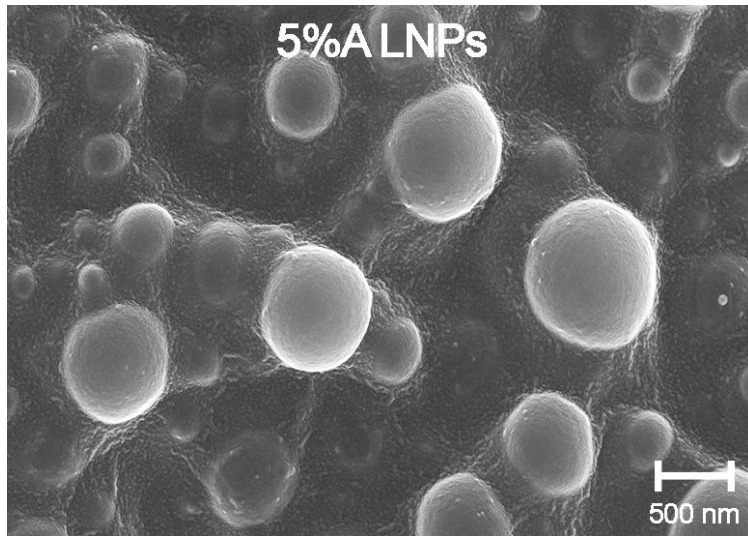
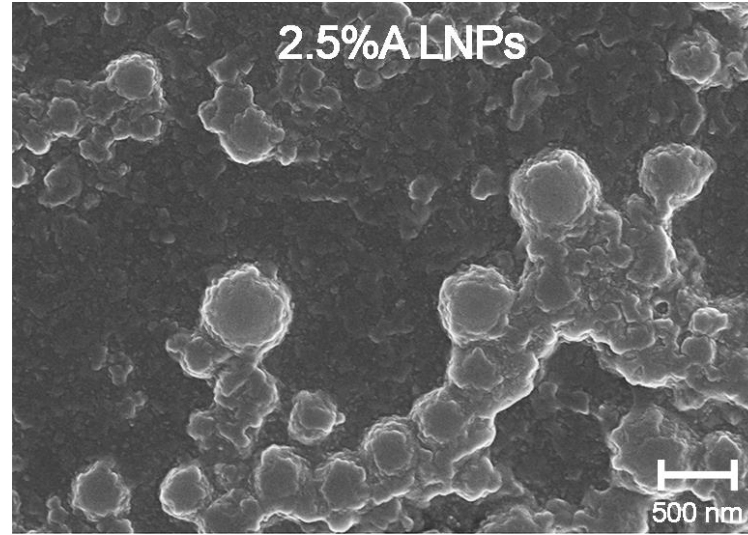
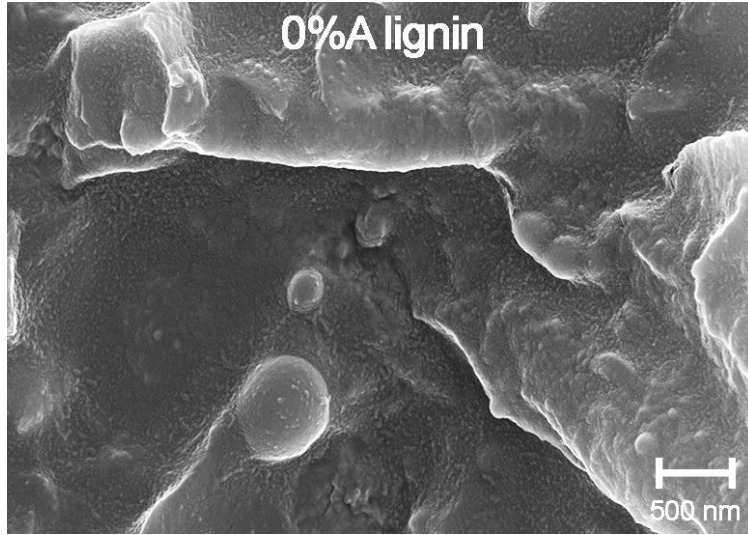
## Solution:

Add a small % of solvent to allow for complete dissolution and homogenous nucleation





# scH<sub>2</sub>O-Produced LNPs



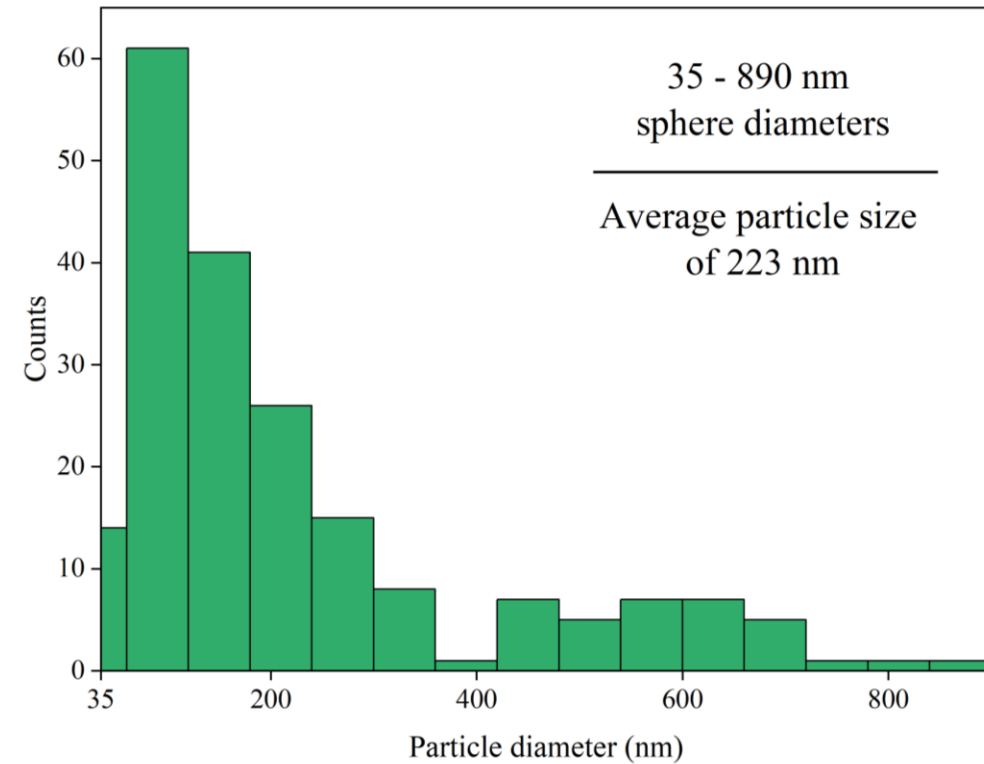
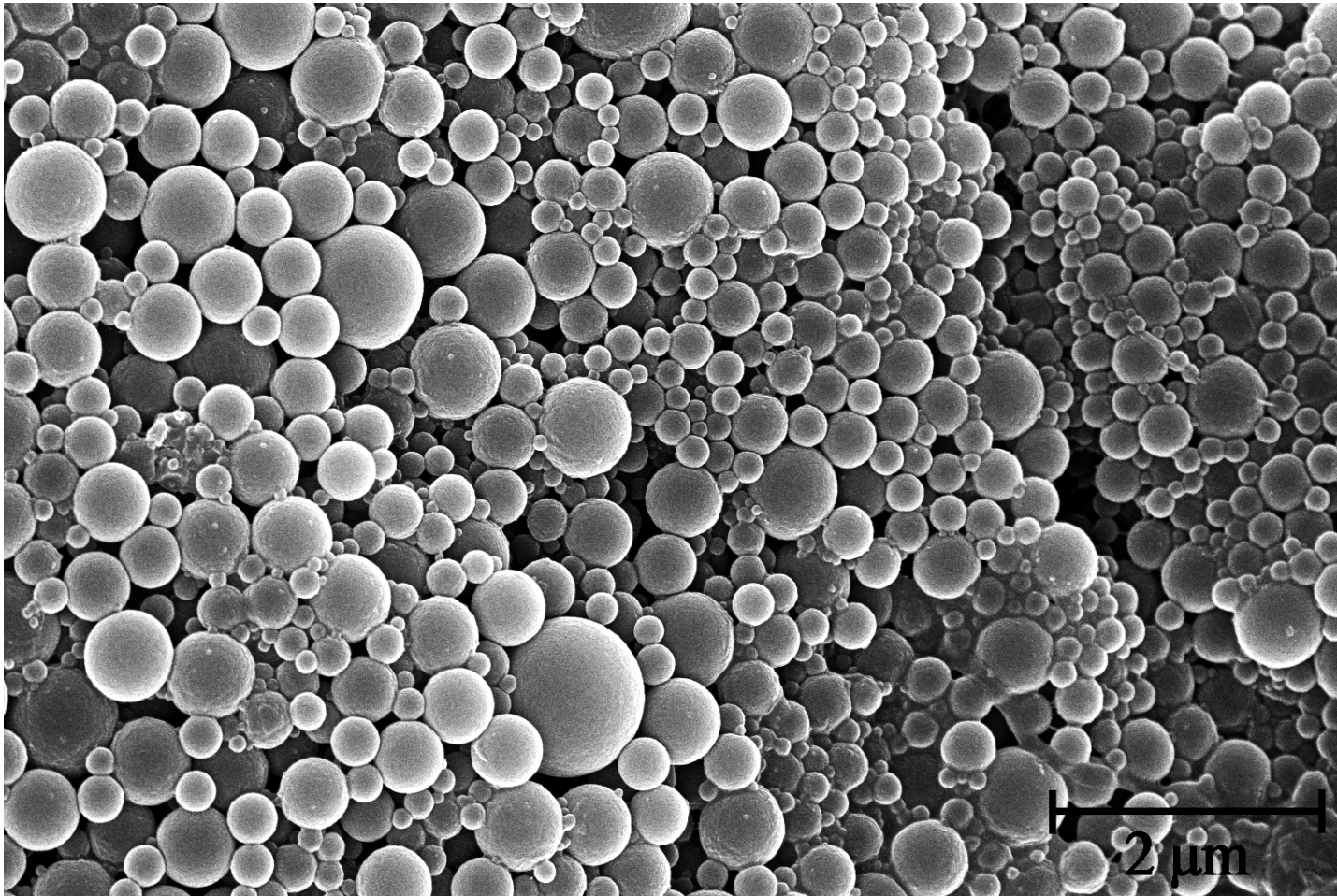
Spherical LNPs were produced, showing that the acetone/subcritical H<sub>2</sub>O treatment had a synergistic effect





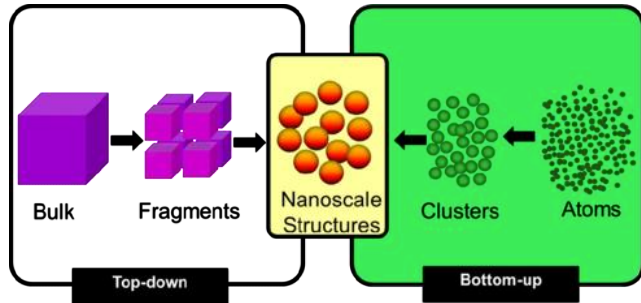
# LNP Morphology

Further dilution revealed well-defined lignin nanoparticles

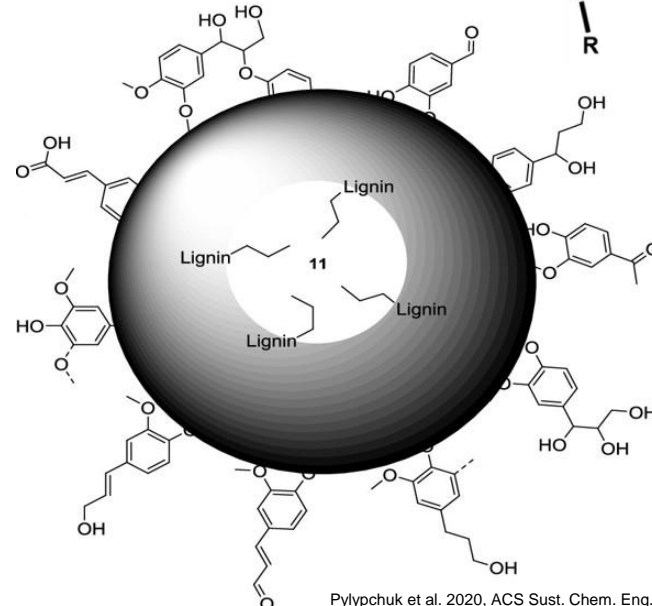
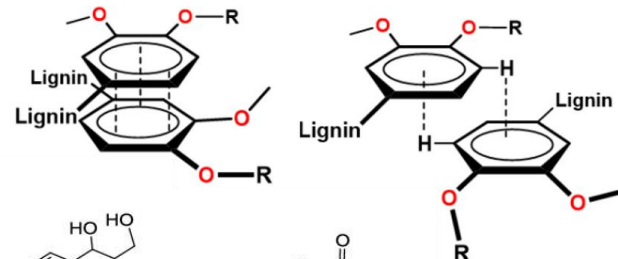


# LNP Morphology

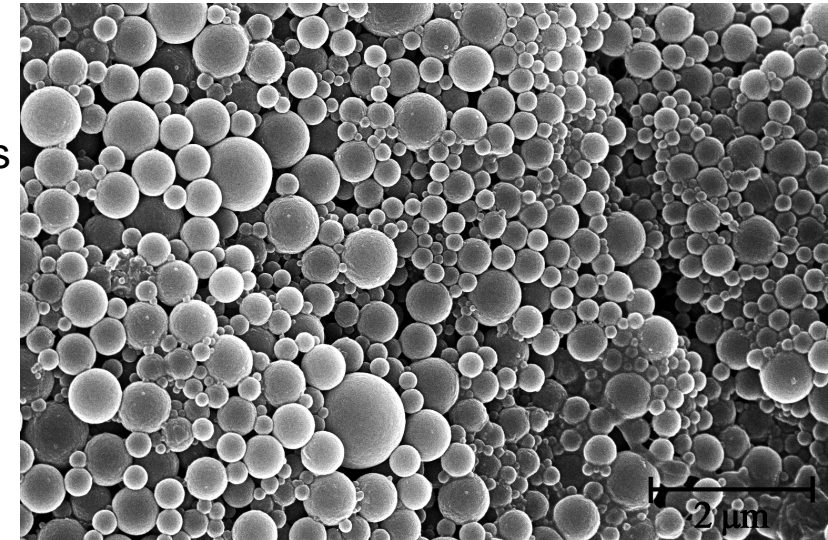
## Nanosphere mechanism of formation



- $\pi$ - $\pi$  stacking dominates, due to the high concentration of flat aromatic ring structures



Pylypchuk et al. 2020, ACS Sust. Chem. Eng.



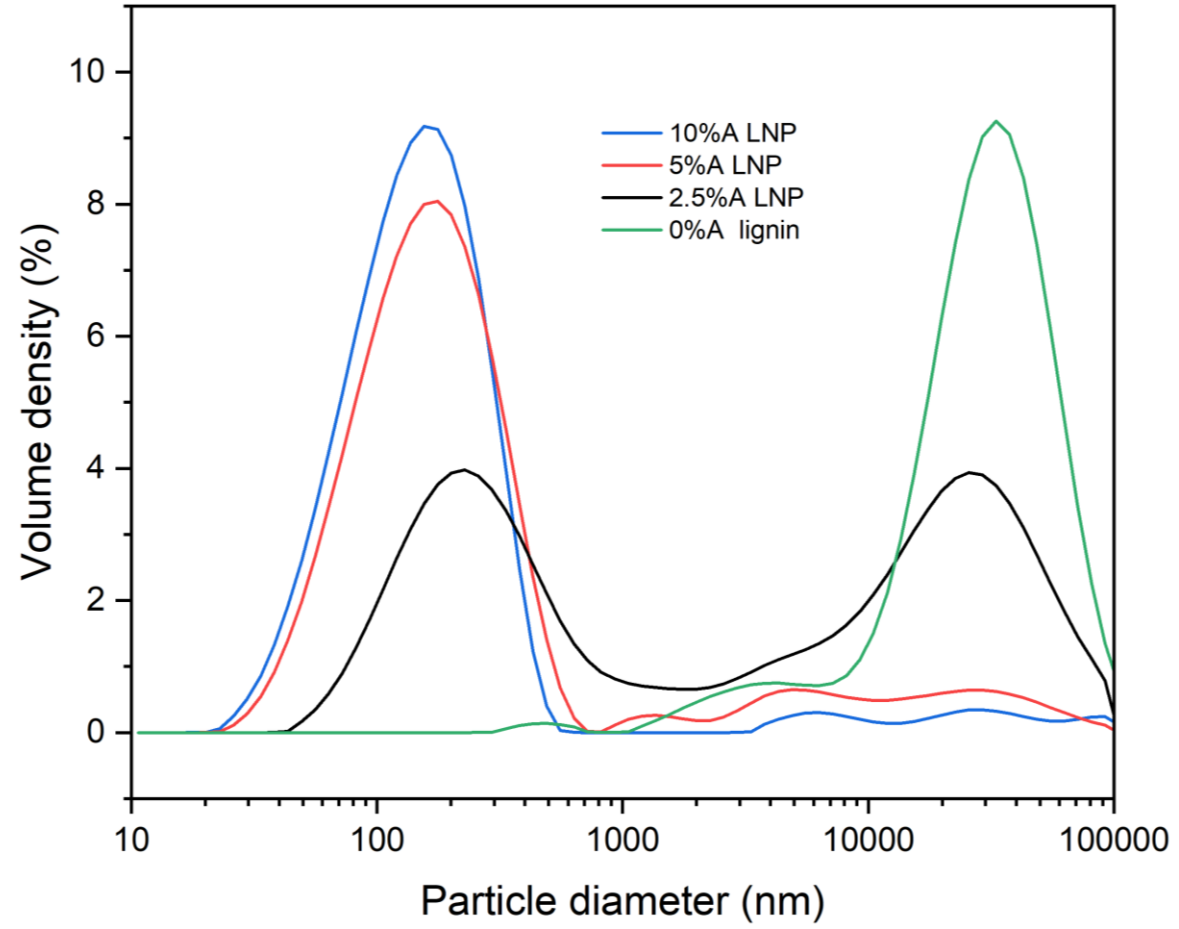
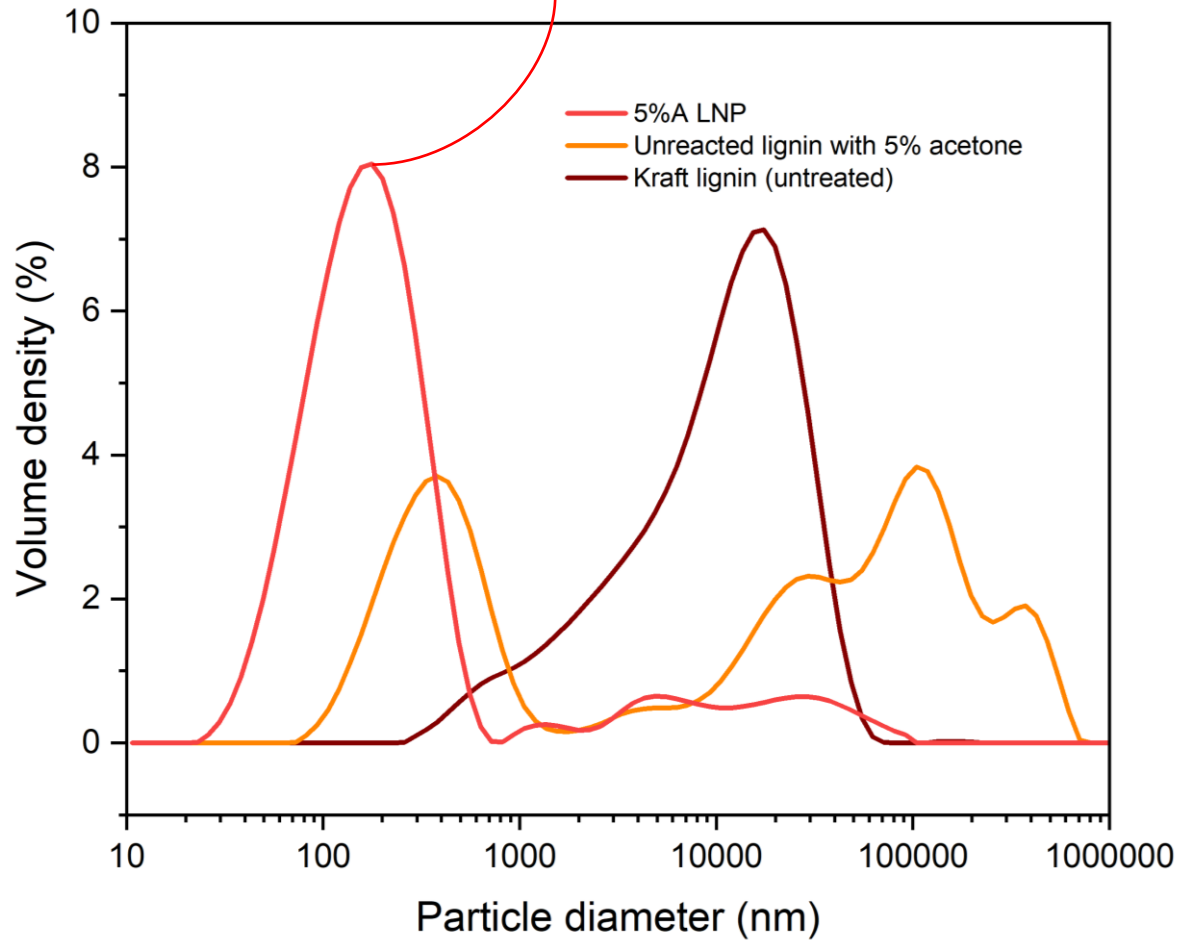
- Initial nucleation (High Mw, aromatic ring stacking) forms core of nuclei
- More hydrophilic, charged units absorb onto the nuclei

- Ionized surface molecules provide electrostatic stability in solution
- Bottom-up self-assembly produces **more uniform LNPs**



# LNP Particle Size Distribution

Nanoscale particle yield\* greatly increased with subcritical treatment

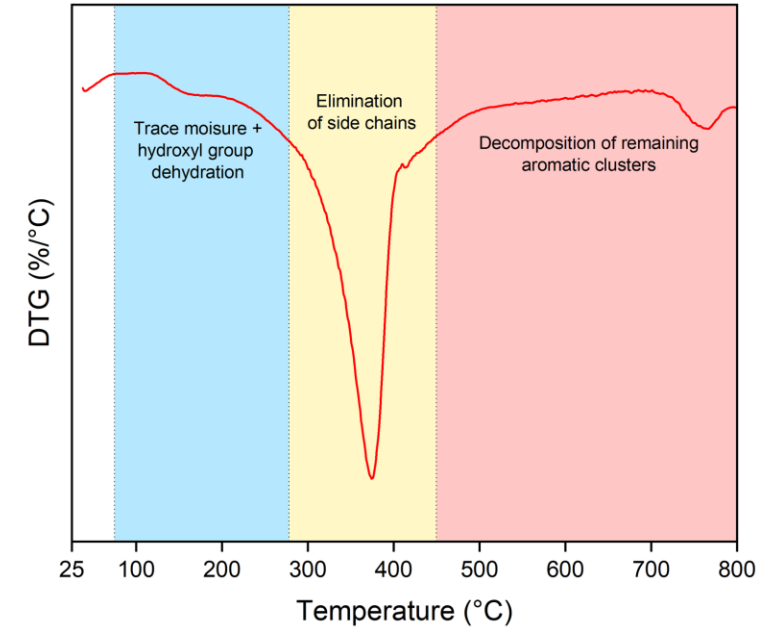
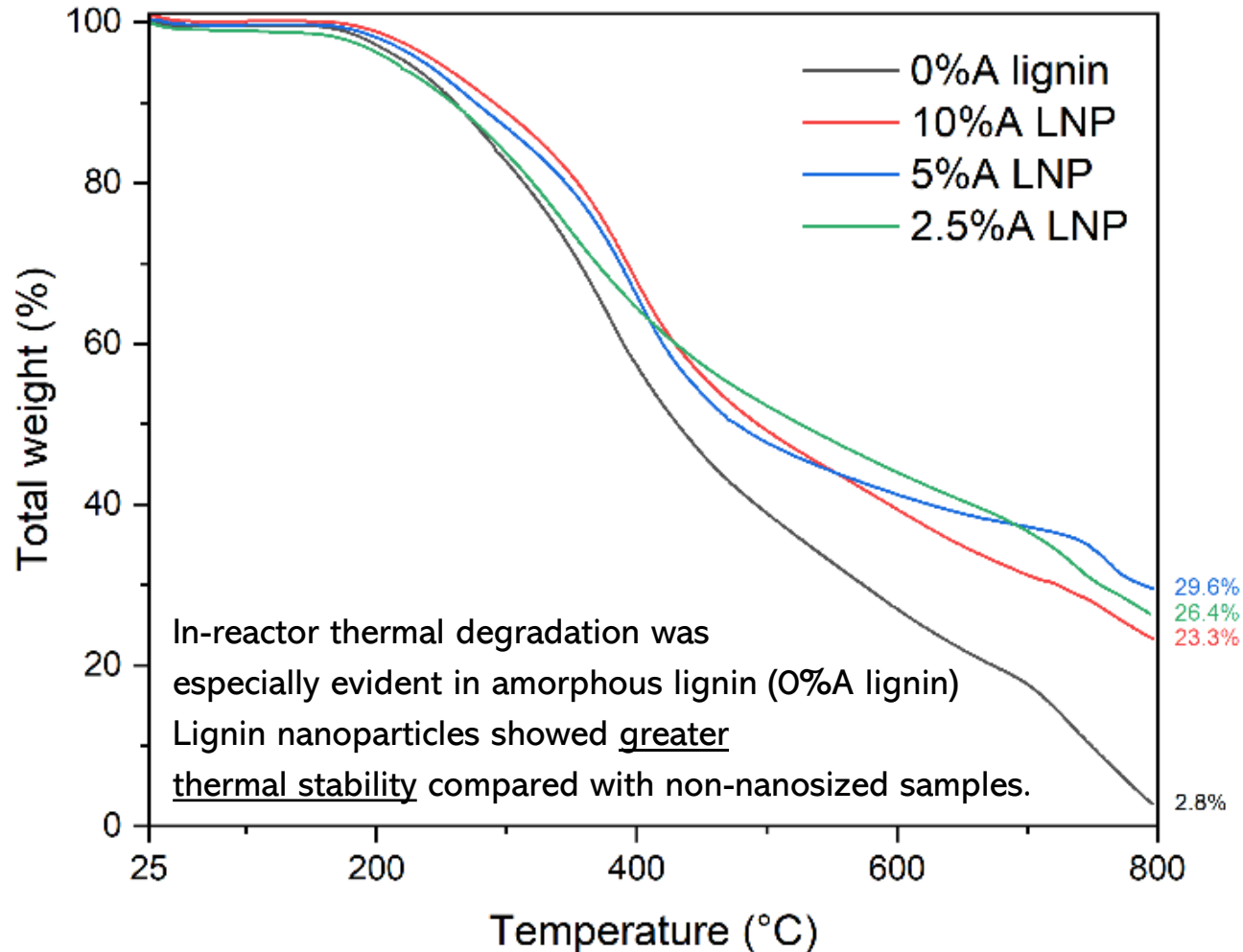


Smaller particle size was directly correlated with acetone concentration

\*Measured with Mastersizer 3000 laser diffraction system



# Thermal Stability of scH<sub>2</sub>O-Produced LNPs



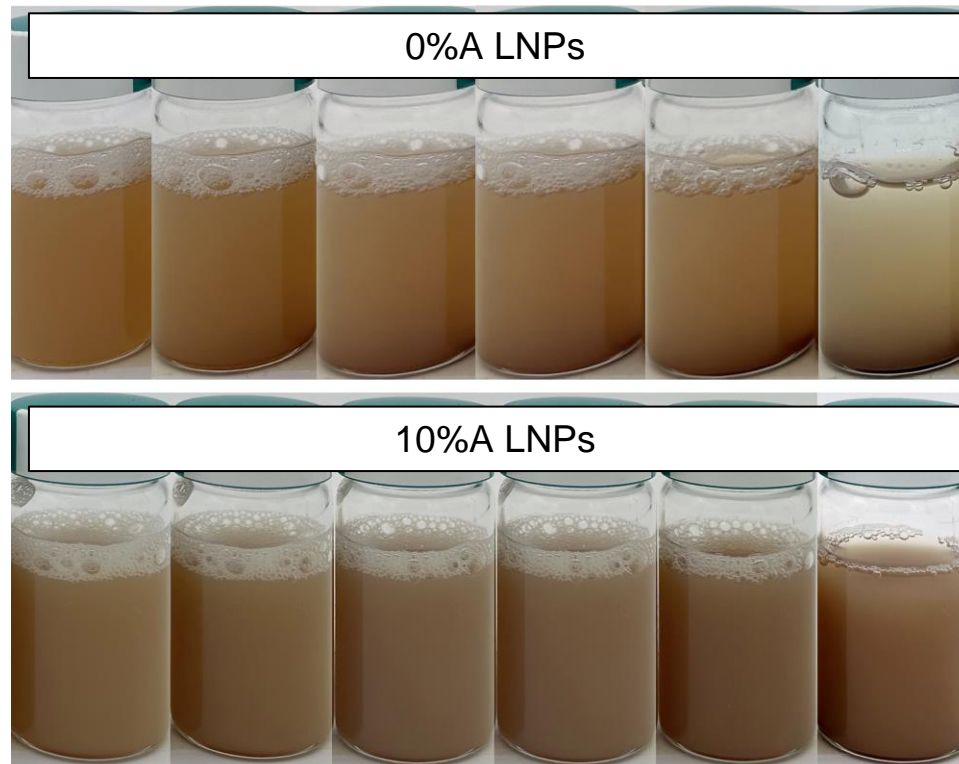
| Sample     | T <sub>onset</sub> * |
|------------|----------------------|
| 0%A lignin | 240.2 °C             |
| 2.5%A LNP  | 233.5 °C             |
| 5%A LNP    | 253.5 °C             |
| 10%A LNP   | 265.5 °C             |

\*: T<sub>onset</sub> calculated based on Nair et al. 2014

# Results: scH<sub>2</sub>O-Produced LNPs

Colloidal nanoparticle solution stability  
was improved

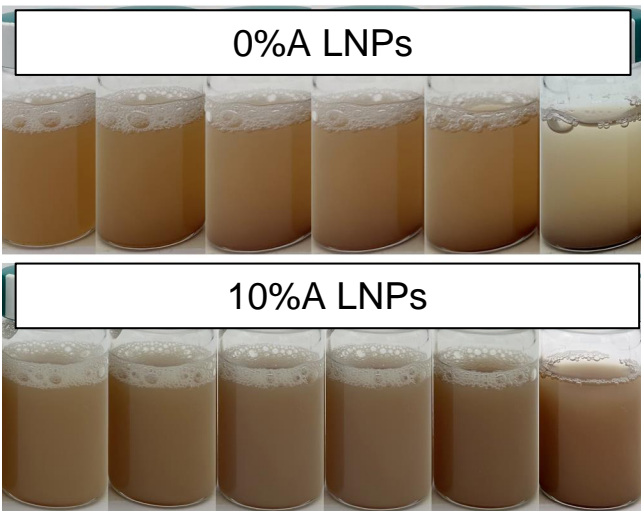
**0 min.   5 min   15 min   30 min   1 hr.   24 hrs.**



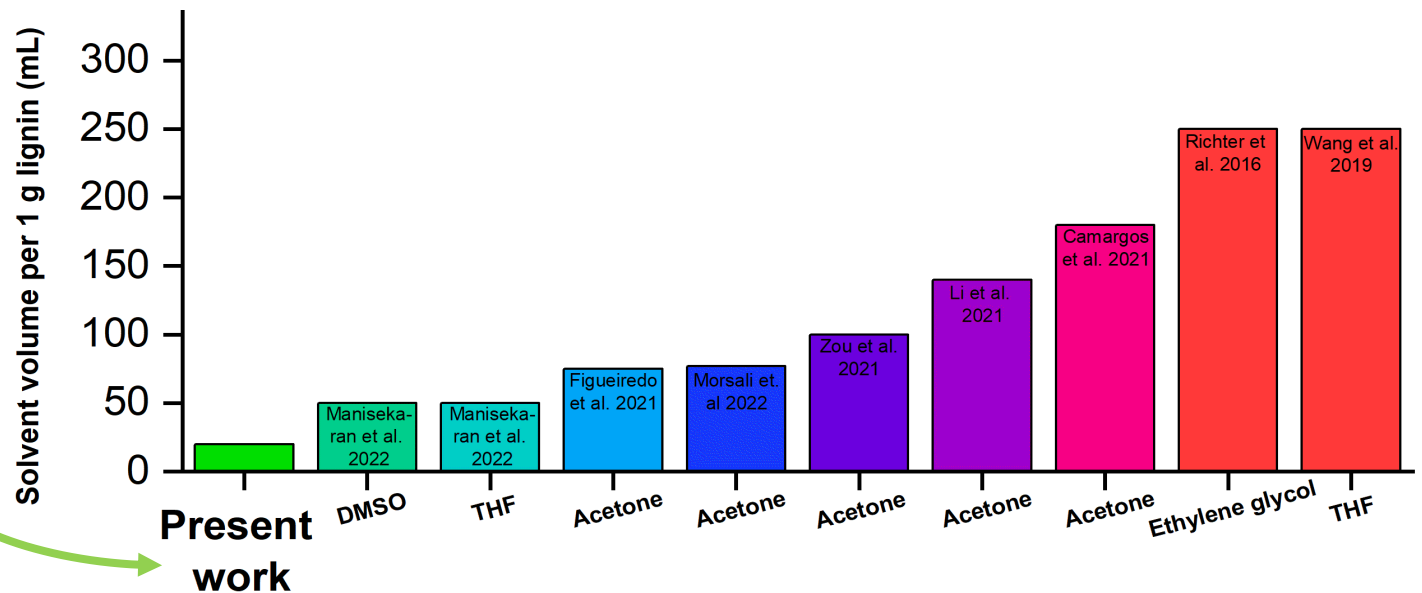
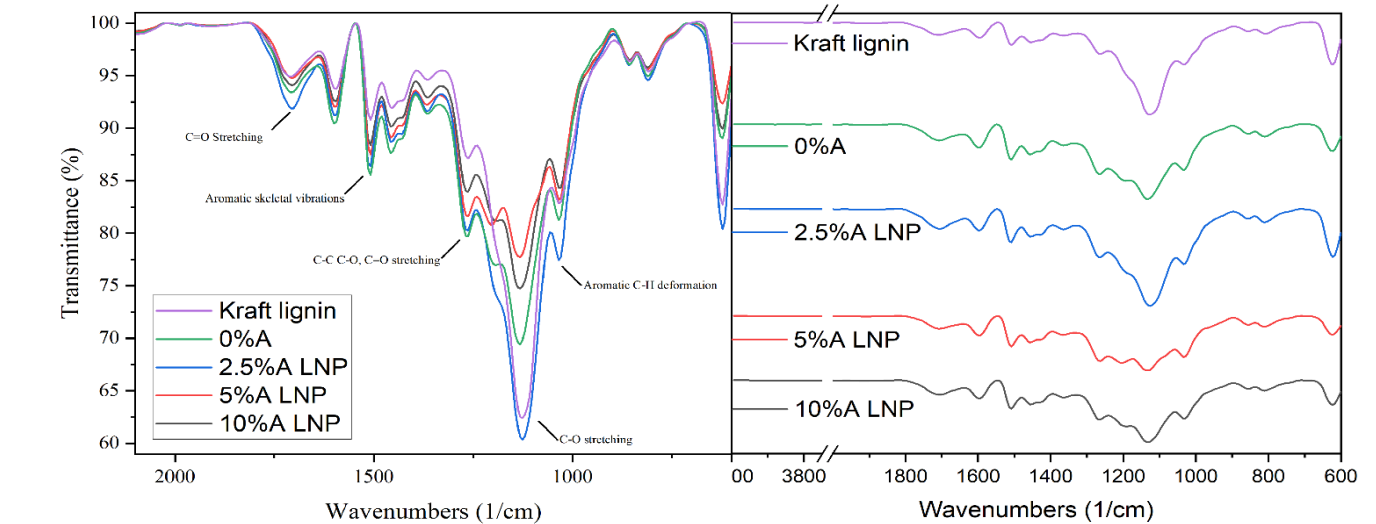
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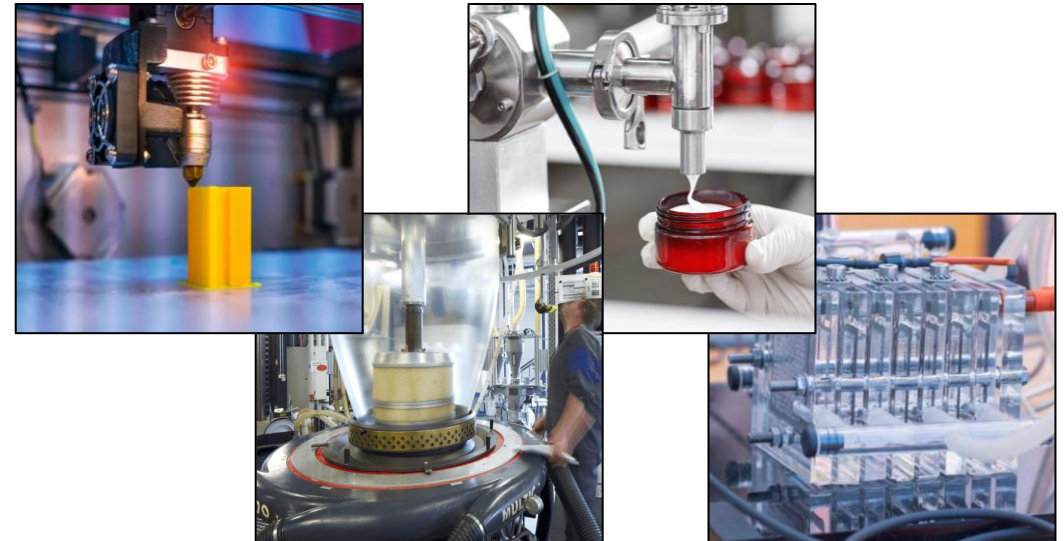
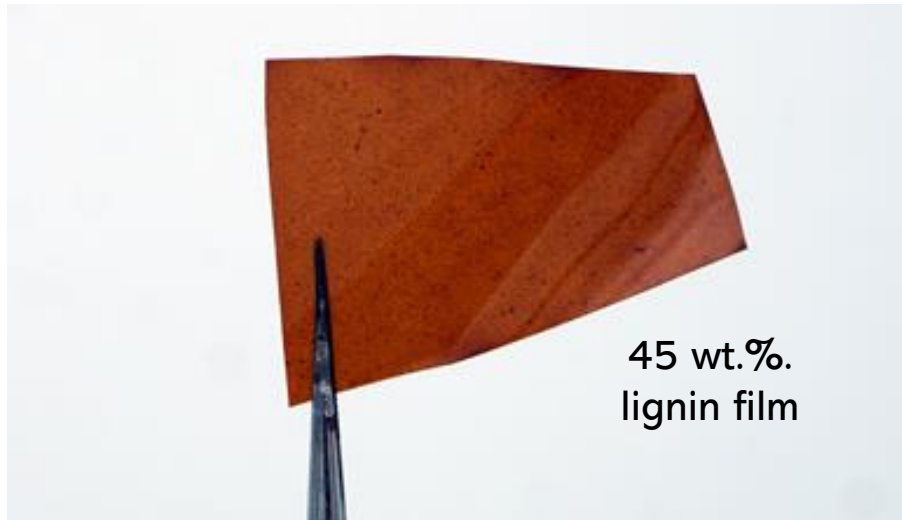


Less solvent volume used vs. prior literature



# Conclusions

- A **novel method** has been presented for lignin nanoparticle synthesis
- Produced LNPs demonstrated **structural uniformity** due to the solvent-antisolvent-initiated self-assembly
- Acetone was used at much **lower volumes** than in prior cases, and may be easily recycled
- This uniform and thermally stable material has potential for a variety of applications...





# Acknowledgements

- Thanks to the entire Foster Advanced Materials Group & Scott Rennecker

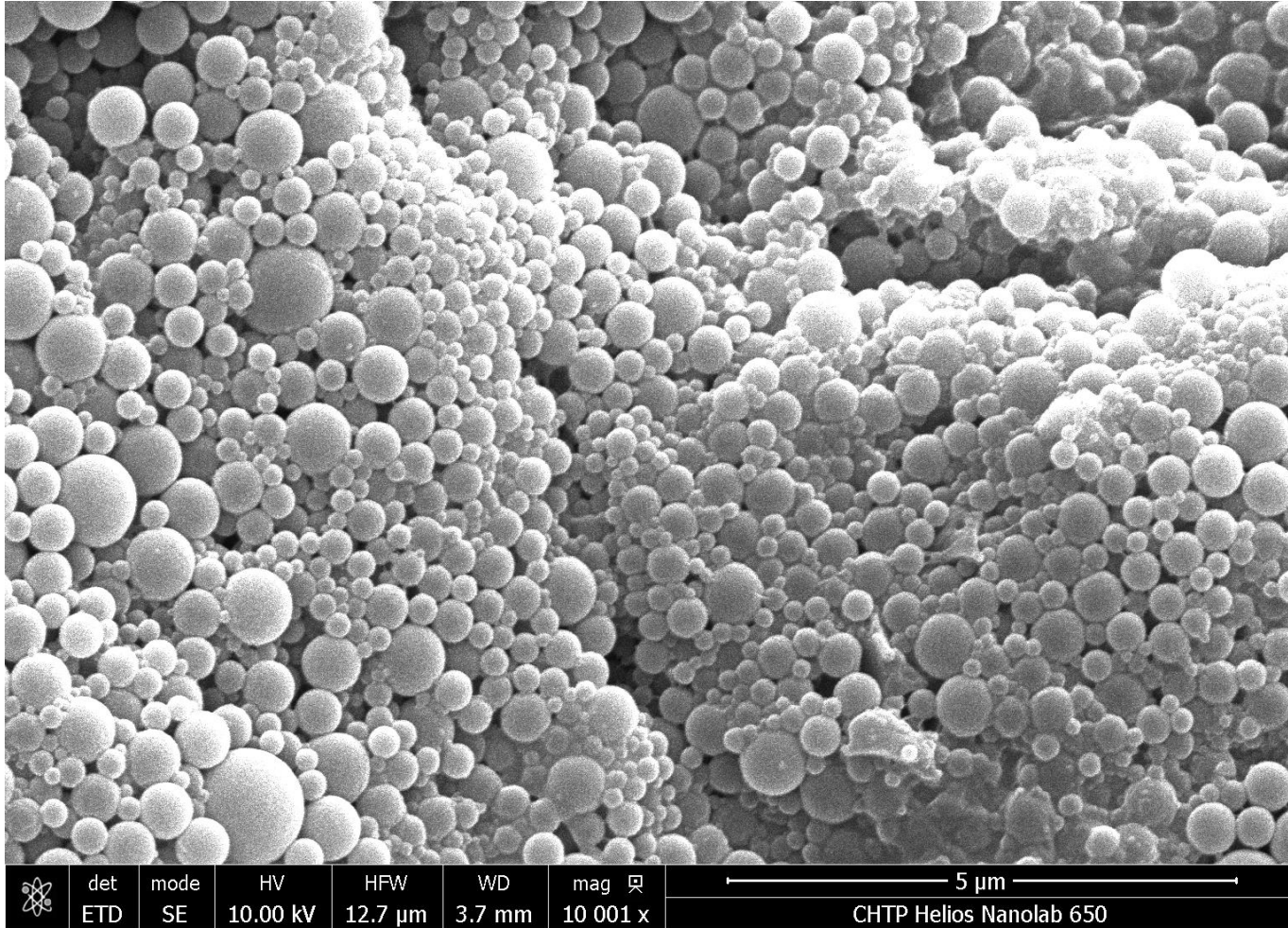


- Special thanks to Victoria French, Fernanda Brito dos Santos, and Behzad Zakani

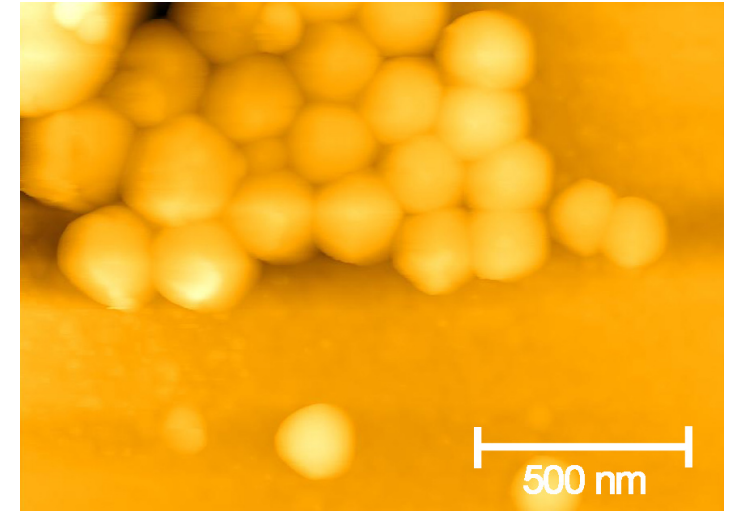
- Thanks to 



# Supplementary Images



SEM imagery by Fernanda Brito



AFM imagery by Ruby Osei-Bonsu

