

Synthesis and Characterization of Microencapsulated Controlled Release Fertilizers (CRFs) by Spray-Drying

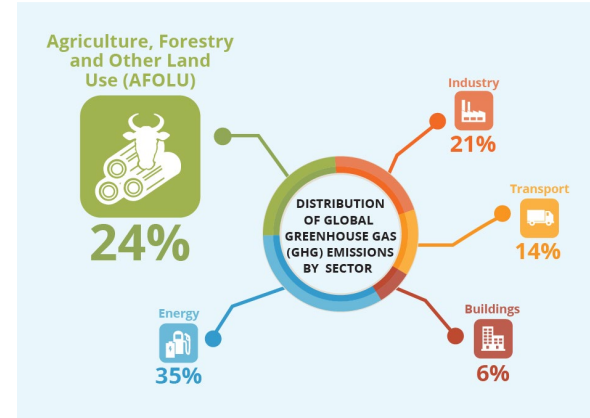
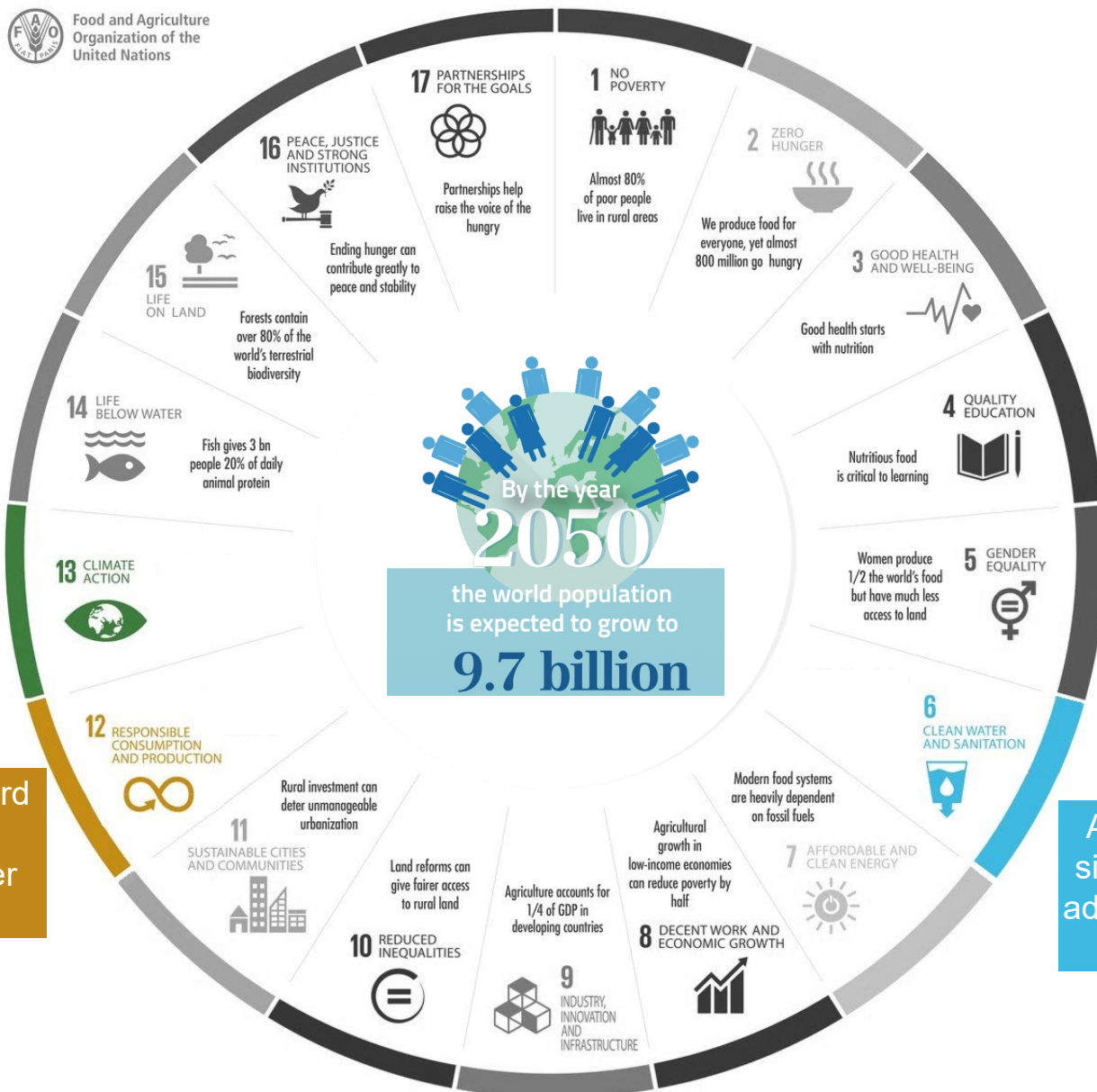
Gutierrez Cesar, Juarez-Luna Gregorio; Rivera-Toledo
Martin, Neri-Torres, Elier, Quevedo Ivan R.



Agenda

- Conventional vs. Controlled-Release Fertilizers (CRFs)
- Objectives of this Investigation
- Experimental Results
- Simulation Model Proposed
- Preliminary Conclusions

The Future of Food and Agriculture



Agriculture plays a substantial role in the exacerbation of global warming

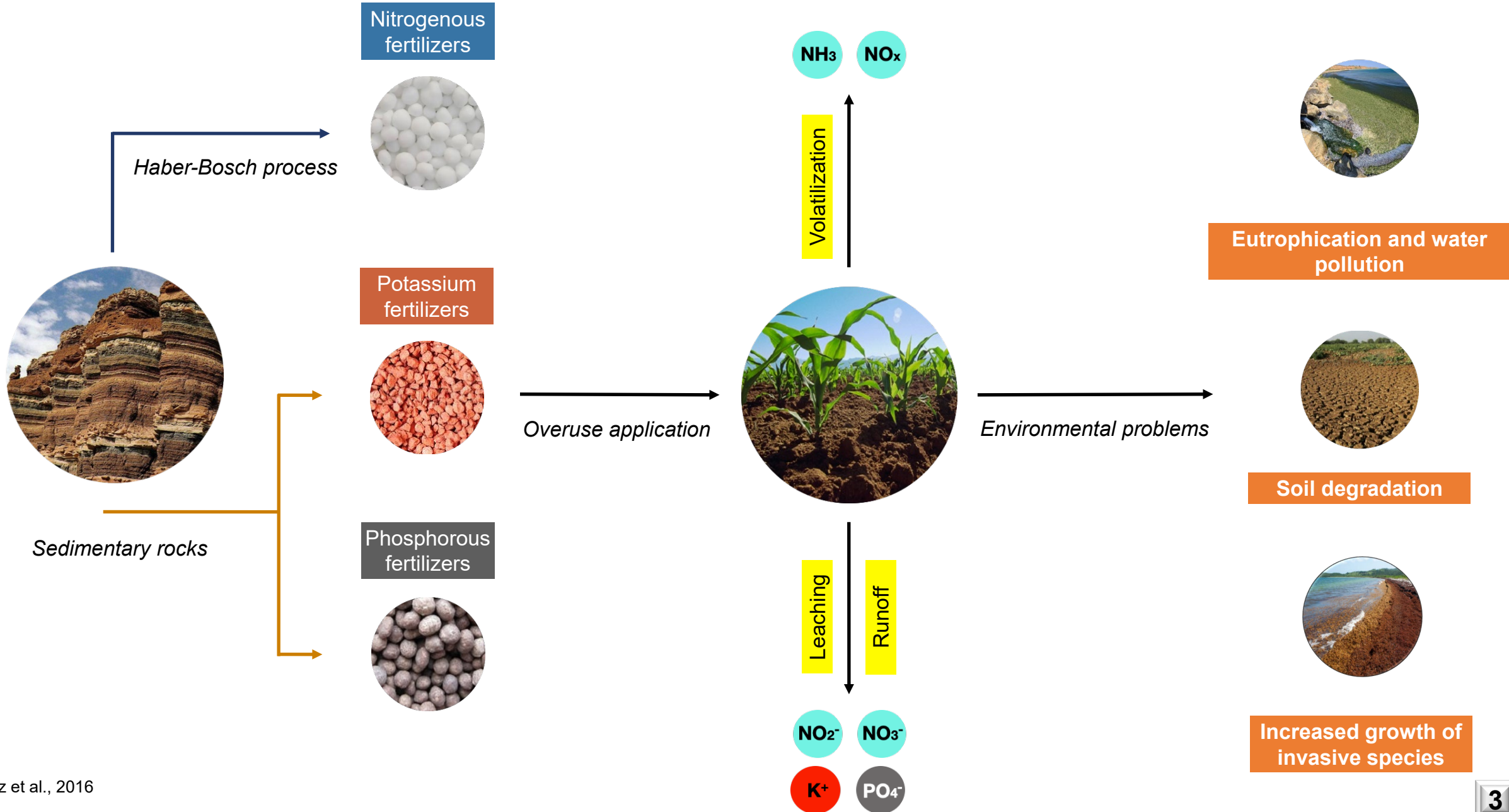


More than one-third of the total food produced is either lost or wasted

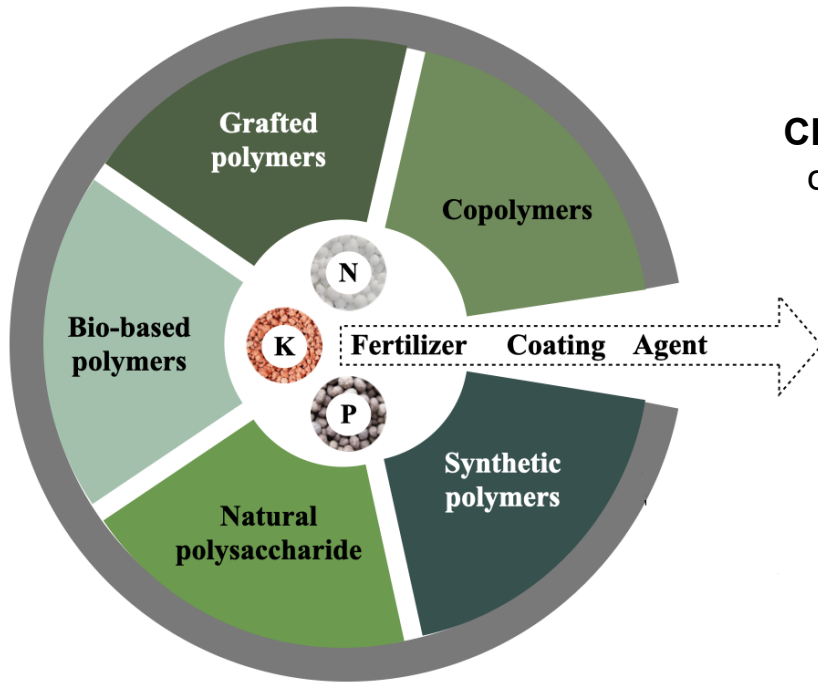


Agriculture presents significant potential in addressing the issue of water scarcity

Conventional Fertilizers

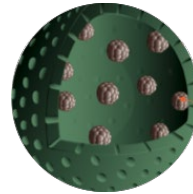


Controlled-Release Fertilizers (CRFs)



CRFs are an alternative to increase crop yield by optimizing **Nutrient Use Efficiency (NUE)**

Biopolymeric matrix system



Biowaste



CRFs



Spray dryer



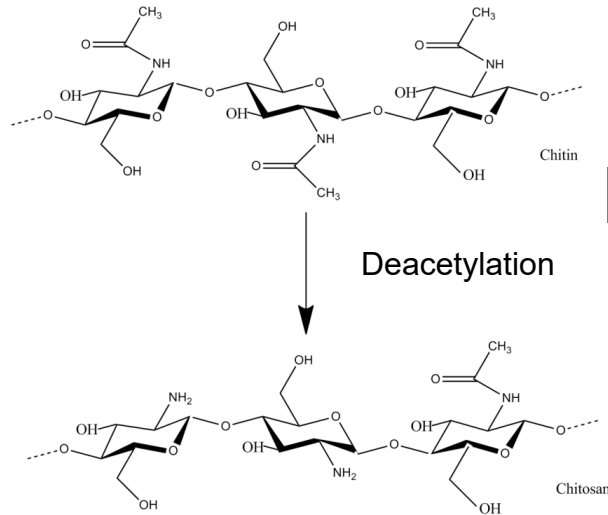
Extracted biopolymers

Lignocellulosic materials	Biopolymers	Biobased polymers
Lignin	Sodium alginate	Biobased polyurethane
Acetylated lignin	k-Carreegeenan	Polyvinyl alcohol
Cotton stalk	Starch	Biobased epoxy
Wheat straw	Chitosan	Poly-eugenol sulfone
Cellulose	Maltodextrin	Epoxy resin
Ethyl cellulose	Sodium alginate	Polyhydroxyalkanoates
Carboxymethyl cellulose	Gelatin	Polylactic acid

CRF Based on Chitosan



Food waste



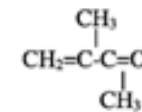
Chitin

Deacetylation

Chitosan

+

MMA

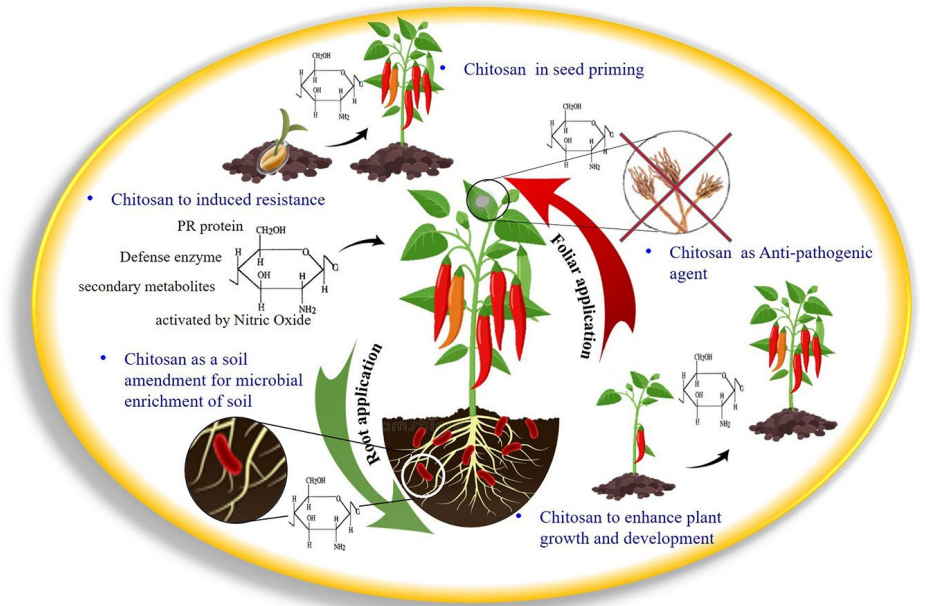


Co-polymerization
chitosan-MMA



+

Urea



Chitosan is a promising encapsulating agent for CRF due to its biocompatibility, lack of toxicity, antibacterial activity, high availability, and natural compound to control plant diseases.

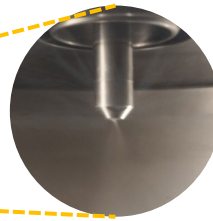
Objectives of this Investigation

- To synthesize a CRF by employing chitosan, MMA (methyl methacrylate), and urea through the process of spray drying. This objective focuses on the production of a fertilizer that releases nutrients gradually over time.
- Modelling the spray drying process to obtain controlled-release fertilizers, enabling the prediction of particle properties and thus enhancing the product's quality.



Spray Dryer (MM-PSR GEA-NIRO)

Inlet temperature
(140 ° C)



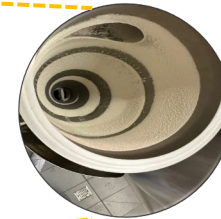
Feed rate
(3 L/h)



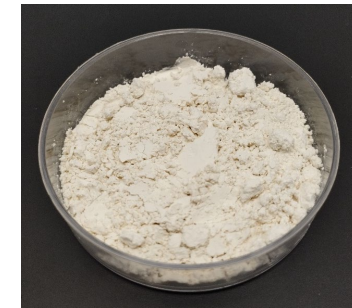
Solution temperature
(22 ° C)

Peristaltic pump

Biopolymeric solution



Atomized pressure
(1.5 bar)



CRF

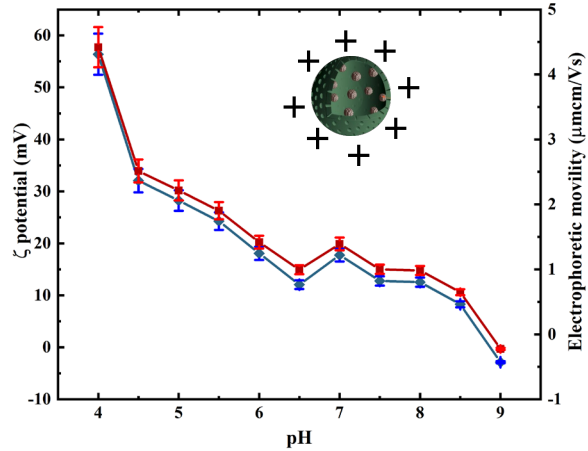
Yield
25-75 %

Outlet temperature
(73 ° C)

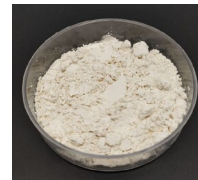
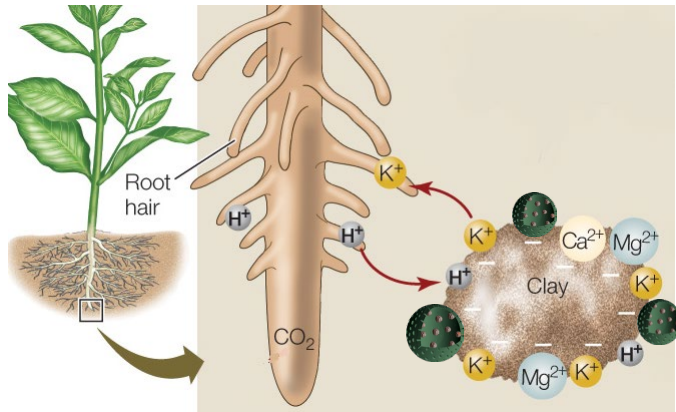
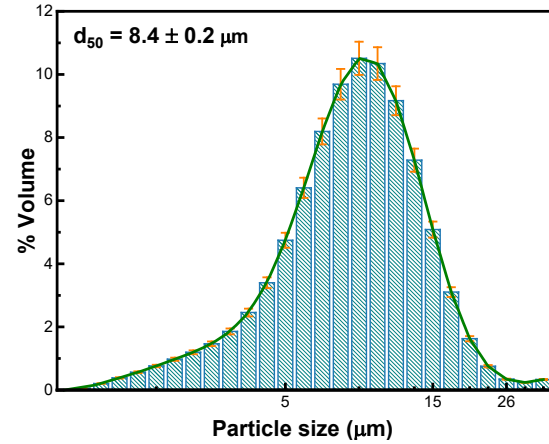


Physicochemical Properties

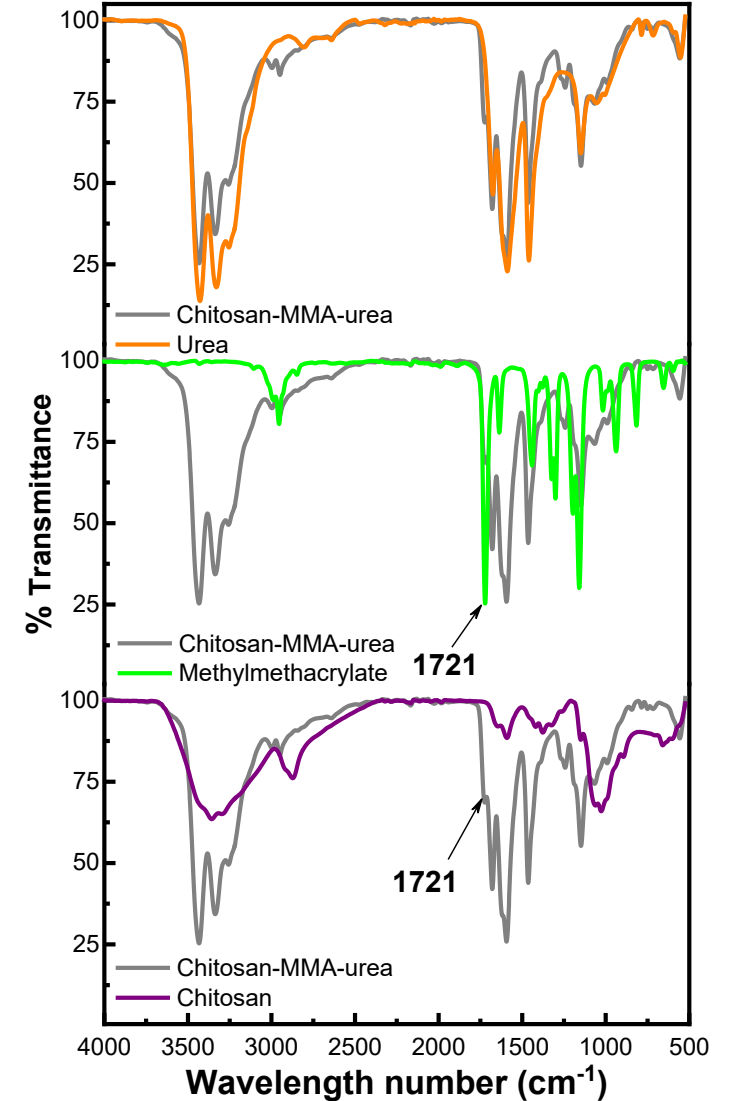
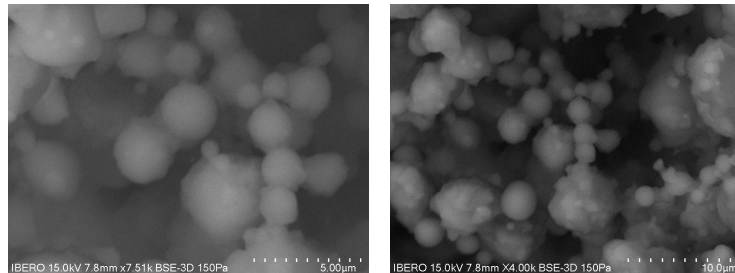
Net surface charge



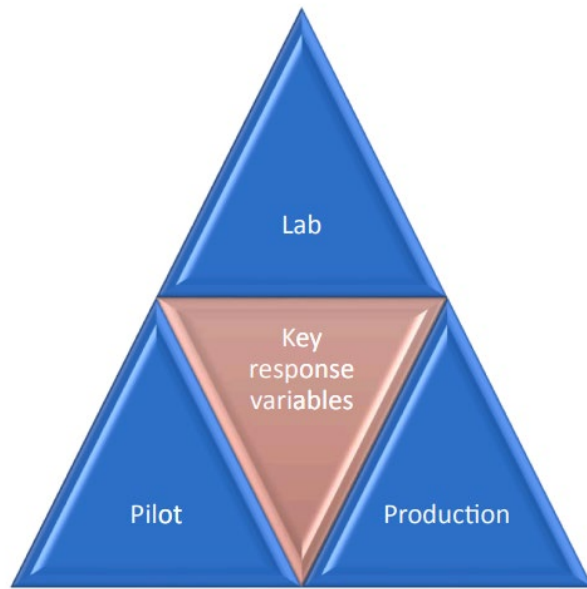
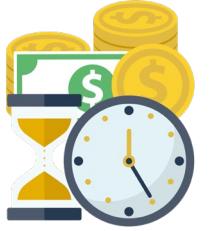
Particle size



Morphology



Simulation of Spray Dryer Performance



Simulation

Scalability process



Mini spray dryer



Pilot spray dryer



Industrial spray dryer

Operational parameters

Process Parameters

Nozzle velocity (a) 38.5 m/s

Drying gas velocity (b) 24.8 m/s

Outlet gas velocity (c) 29.7 m/s

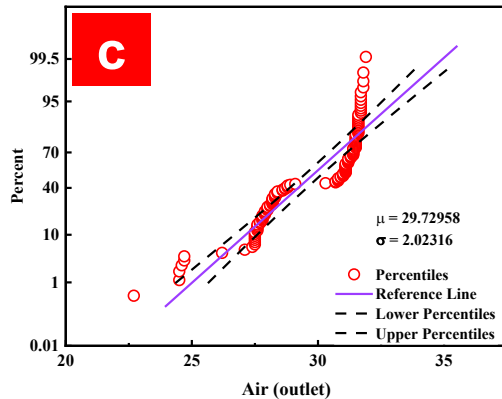
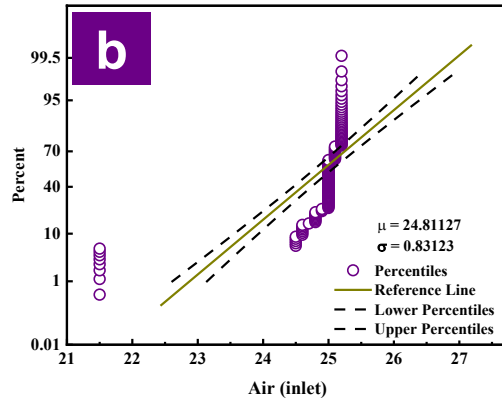
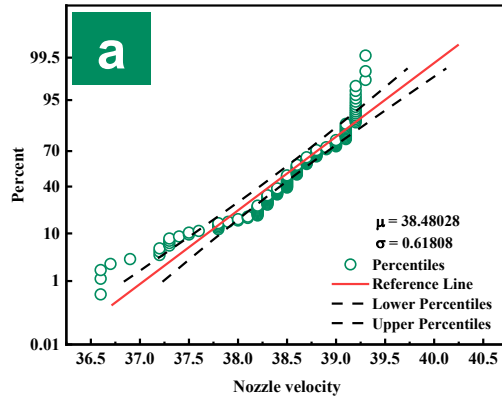
Response Variables

Particle mass, size, and velocity

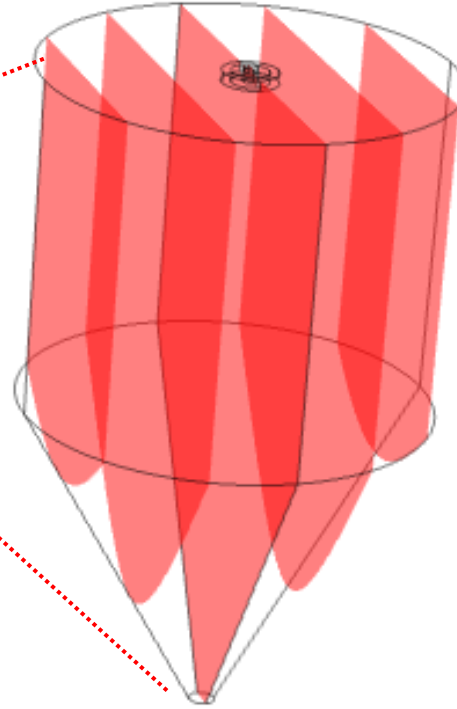
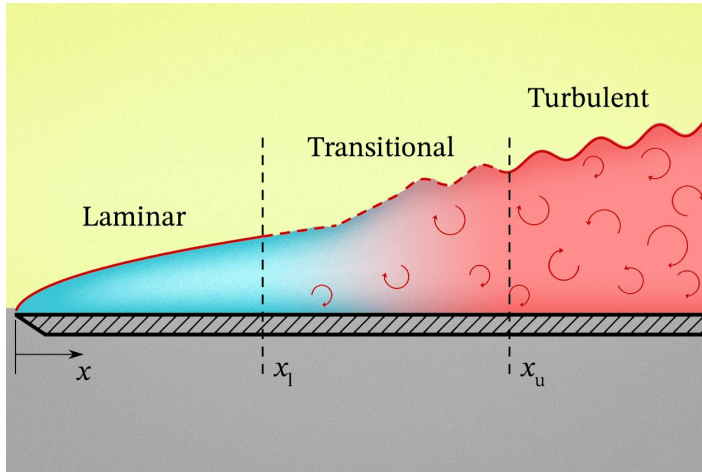
Humidity content

Outlet temperature

Velocity profile (drying gas)



Turbulent flow



Computational Fluid Dynamics (CFD)

k - ϵ model

(widely employed in industrial processes)

Turbulent kinetic energy (k)

$$\rho \frac{\partial k}{\partial t} + \rho u \cdot \nabla k = \nabla \cdot \left(\left(\mu + \frac{\mu_T}{\sigma_k} \right) \nabla k \right) + P_k - \rho \epsilon$$

Turbulent dissipation rate (ϵ)

$$\rho \frac{\partial \epsilon}{\partial t} + \rho u \cdot \nabla \epsilon = \nabla \cdot \left(\left(\mu + \frac{\mu_T}{\sigma_\epsilon} \right) \nabla \epsilon \right) + C_{\epsilon 1} \frac{\epsilon}{k} P_k - C_{\epsilon 2} \rho \frac{\epsilon^2}{k}$$

Reynolds stress

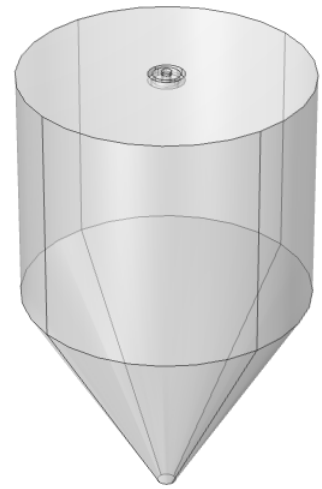
$$\overline{\rho u'v'} = \mu_t \frac{\partial \bar{u}}{\partial y}$$

Turbulent viscosity

$$\mu_T = \rho C_\mu \frac{k^2}{\epsilon}$$

Turbulent viscosity (μ_T) is obtained using de **k - ϵ model**

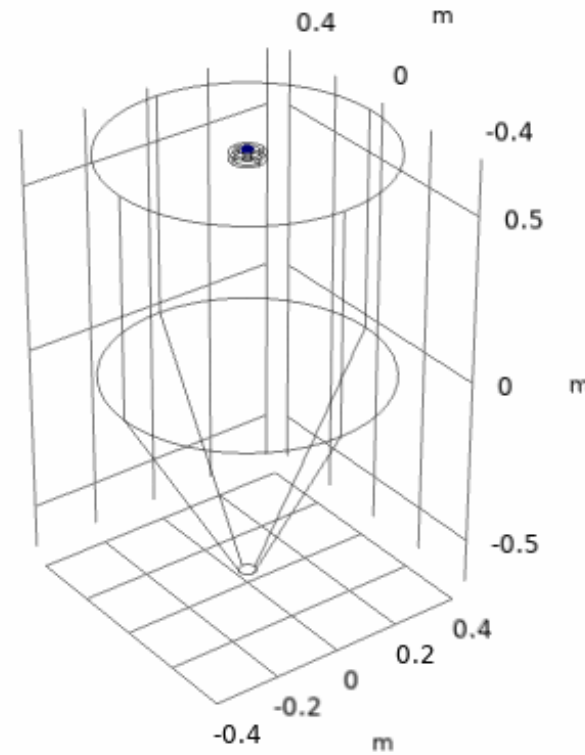
Preliminary results with CFD Modelling



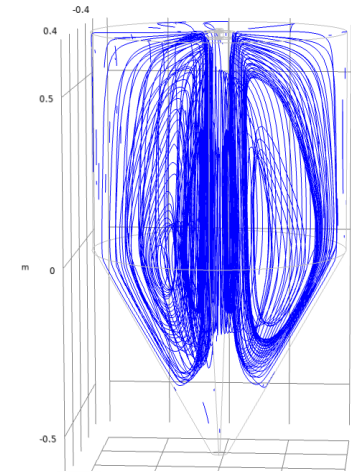
Geometry

Time=0 s

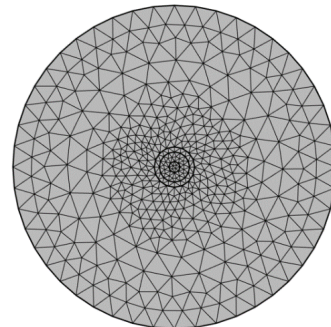
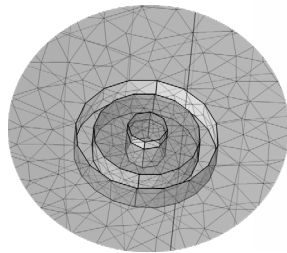
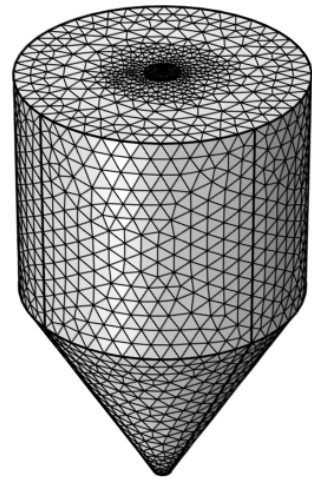
Particle trajectories



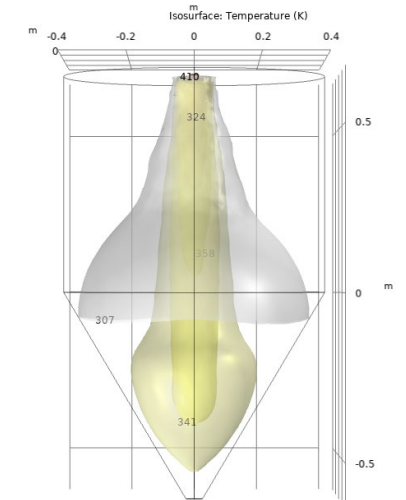
Air velocity field



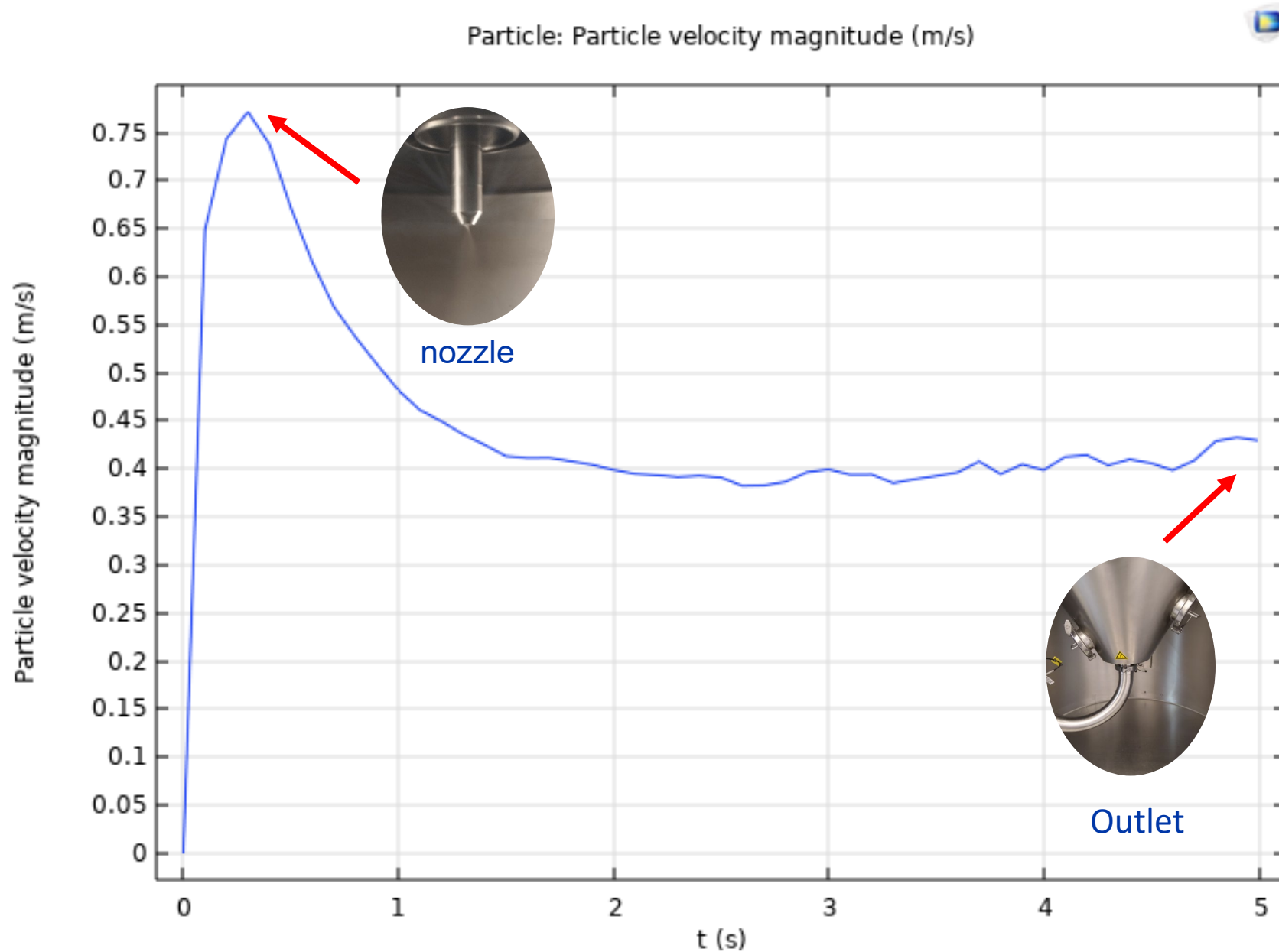
Mesh



Isosurface temperature

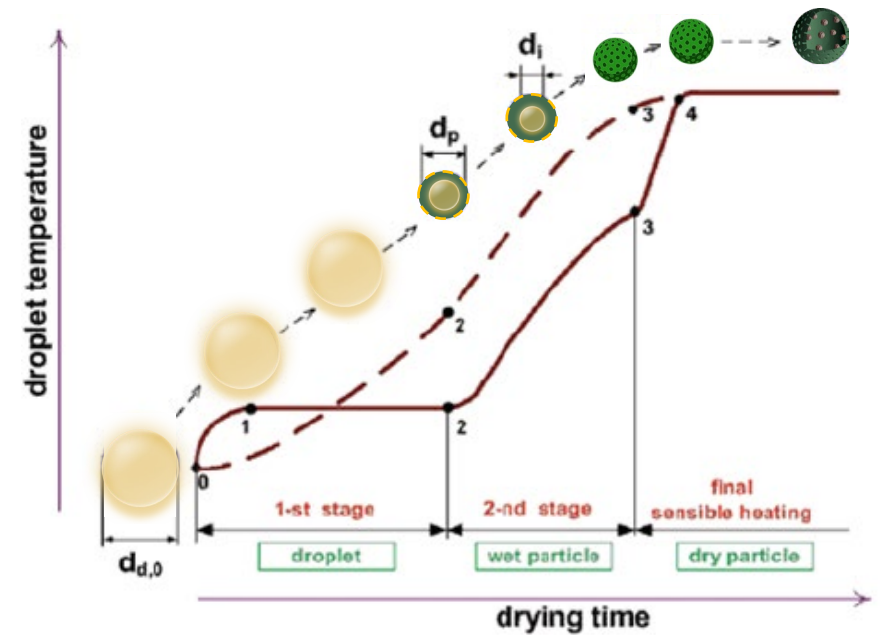
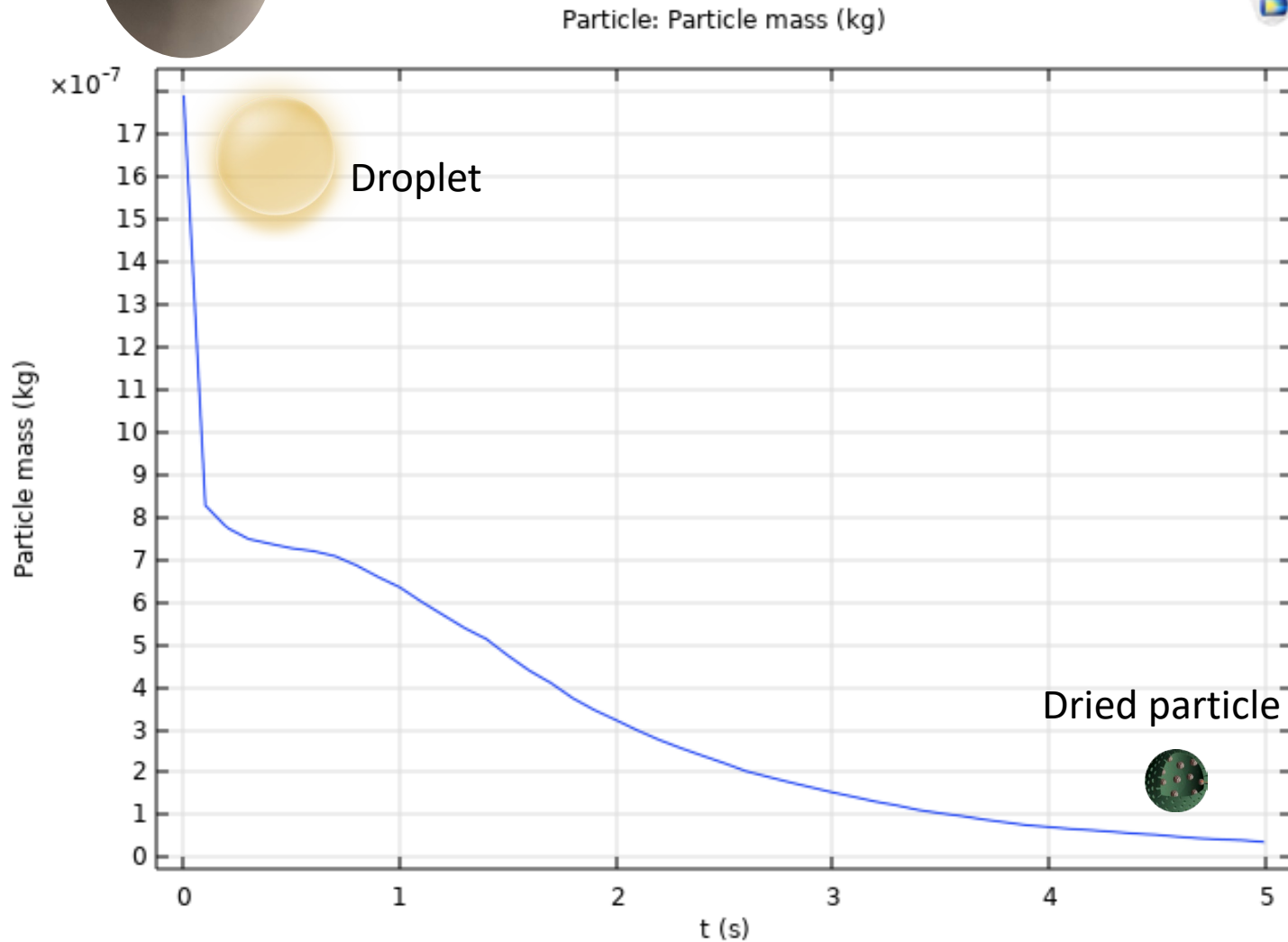


Preliminary results with CFD software





Preliminary results with CFD software



Preliminary Conclusions

- Controlled-release fertilizers (CRFs) derived from biopolymers present a viable solution for implementing sustainable agricultural practices.
- The process of spray drying enables the conversion of controlled-release agrochemicals (like CRFs) into powder form. This conversion is anticipated to enhance storage, transportation, and application procedures of these products in the field.
- Employing Computational Fluid Dynamics (CFD) offers the possibility of accurately predicting the properties of dried CRFs. This technique could potentially assist in scaling up the spray drying process, resulting in more efficient resource utilization.



Thank you



Dr. Martín Rivera



Dr. Elier E. Neri

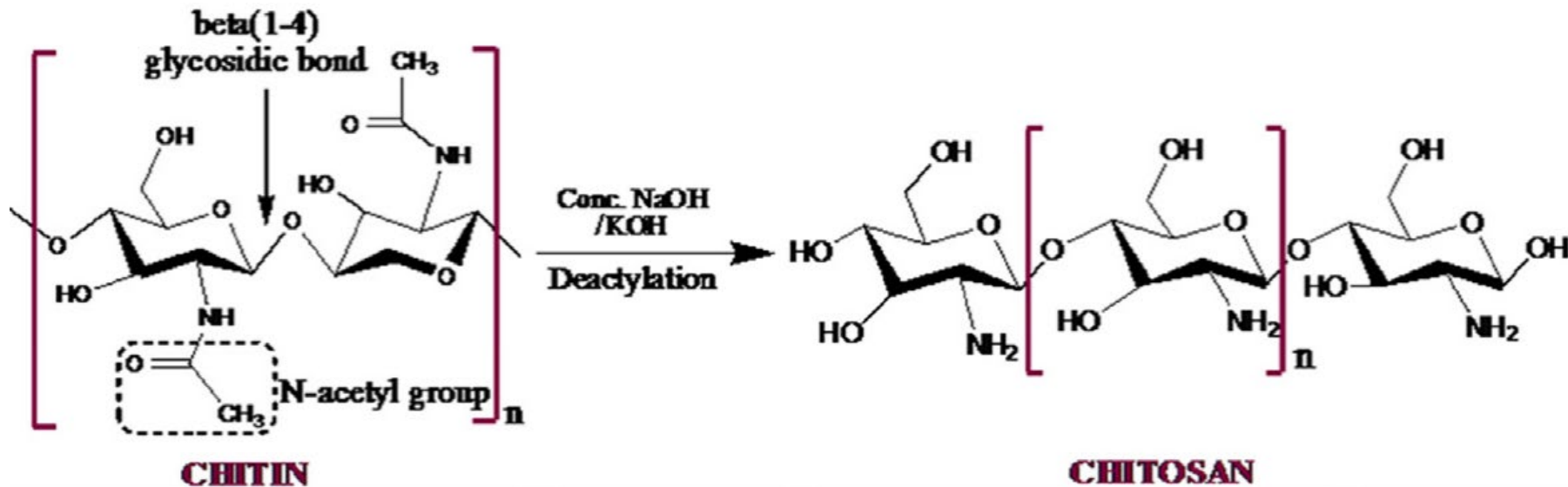


Dr. Gregorio Juárez



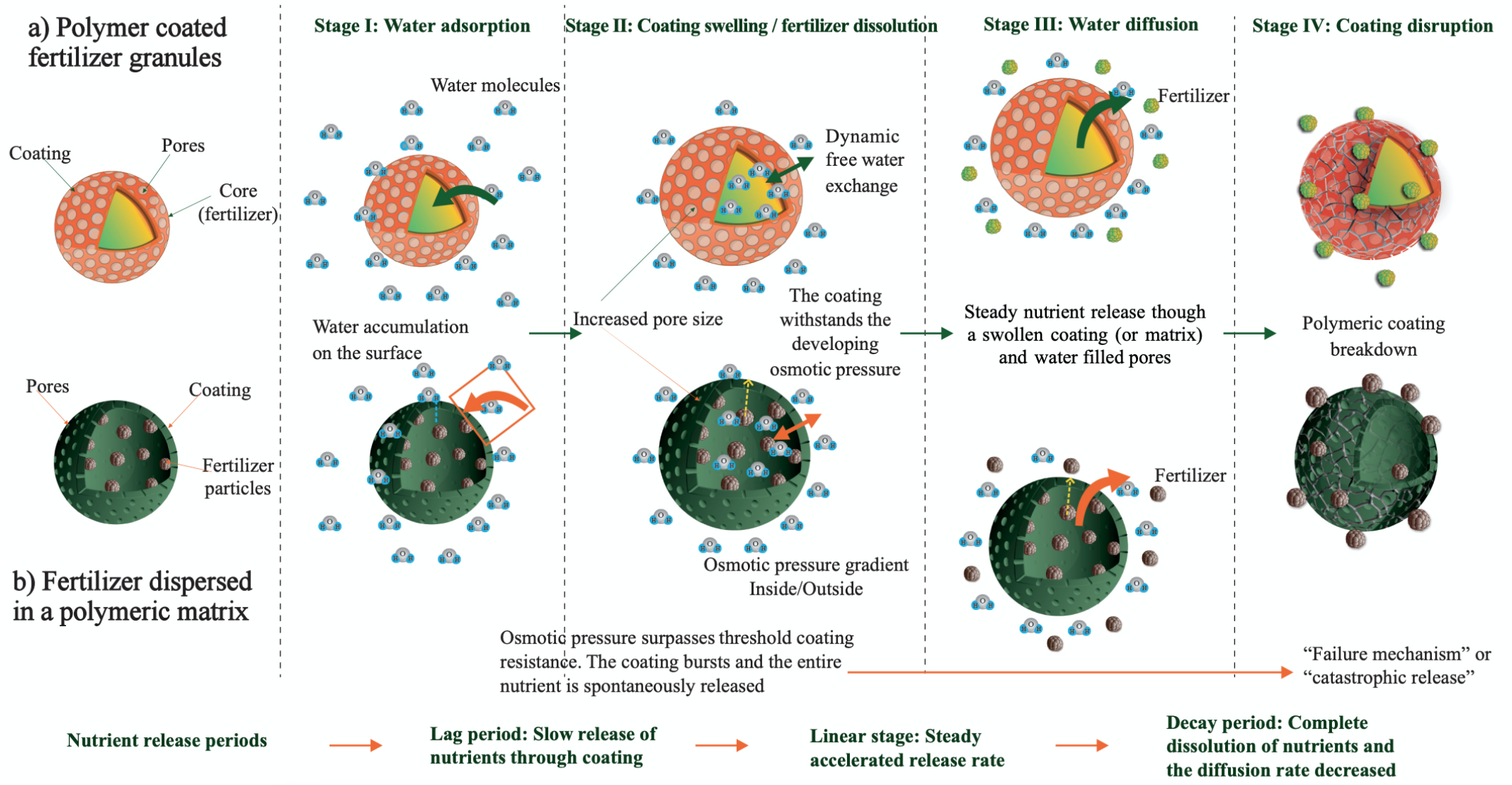
Dr. Iván R. Quevedo

Chemical deacetylation has many disadvantages like high energy consumption and environmental pollution problems. An alternative method of enzyme deacetylation has been developed to overcome these drawbacks.

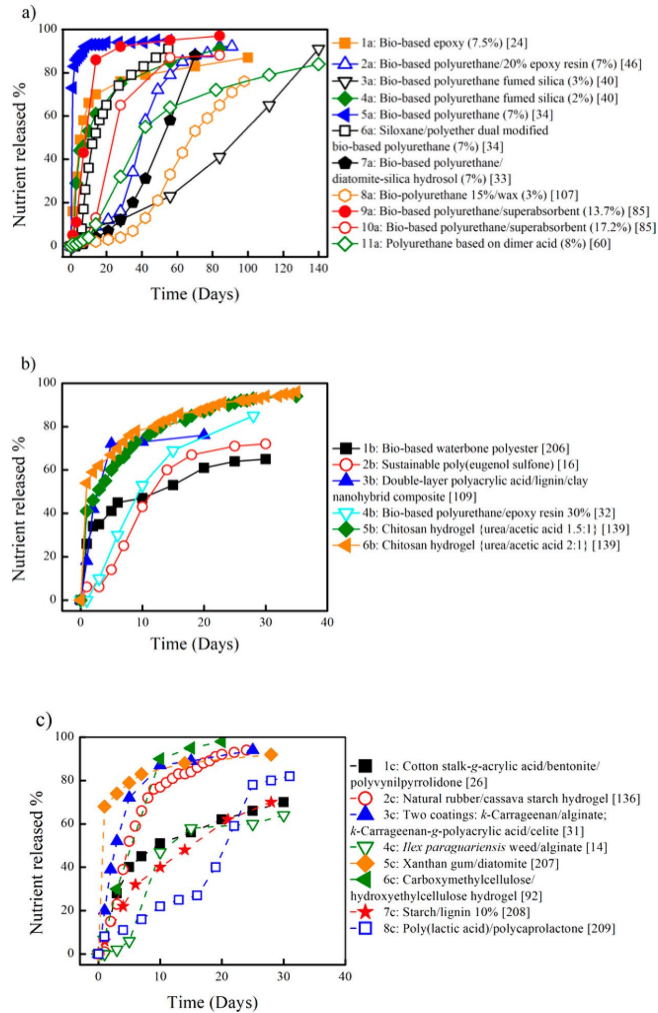


The solubilization occurs by protonation of the -NH₂ group on the C-2 position of the D-glucosamine repeat unit, whereby the polysaccharide is converted to a polyelectrolyte in acidic media

Representing schematics for nutrient release stages in two different CRFs configurations: (a) fertilizer (core) covered by a biobased polymer (coating) to form a granule; and (b) fertilizer particles dispersed into the biobased polymer matrix.



Nutrient Release Behavior in CRFs

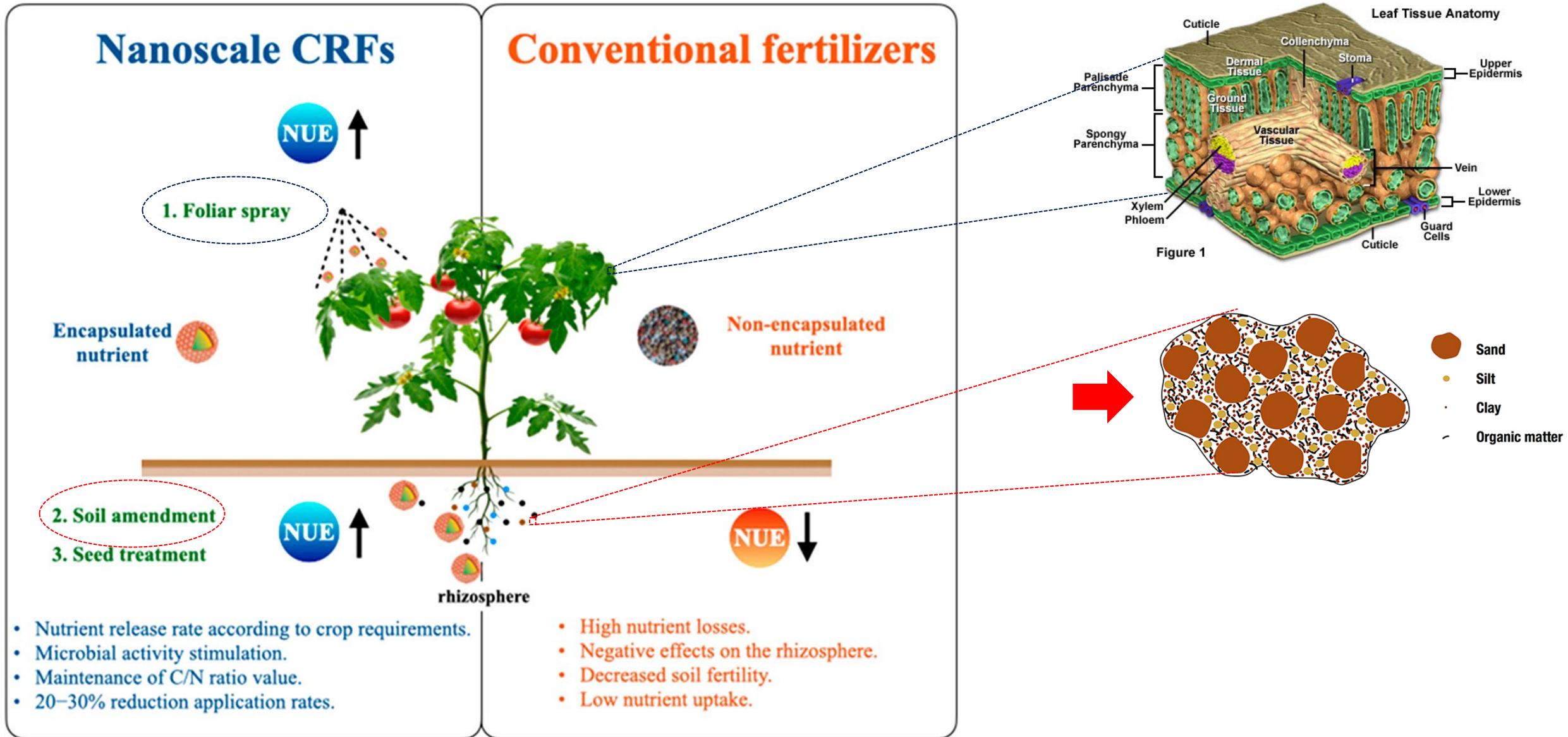


Urea release behavior in water at different time scales in water at (a) 140 days and 35 days (b); in soil at (c) 35 days. In the legend, the coating percentage in the formulation is indicated in parentheses.

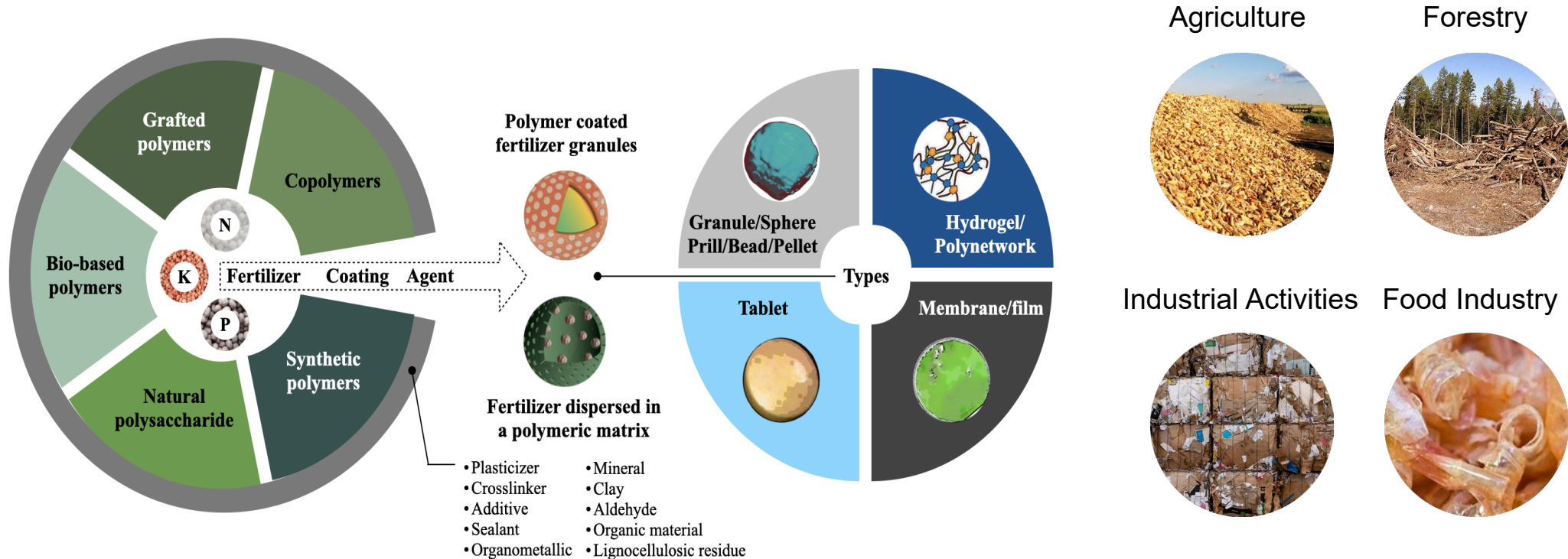
Zeroth-order ¹³¹ $\frac{M_t}{M_\infty} = k_0 t$	Linear ¹³⁸ $M_t = a + bt$	Elovich ¹⁶ $M_t = b + k \ln t$
First-order ¹⁸⁵ $\frac{M_t}{M_\infty} = 1 - e^{-k_1 t}$	Higuchi ¹³¹ $\frac{M_t}{M_\infty} = k_H t^{\frac{1}{2}}$	Hixon-Crowell ⁶⁸ $\left(1 - \frac{M_t}{M_\infty}\right)^{\frac{1}{3}} = 1 - k_H t$
Parabolic diffusion ¹⁶ $M_t = b + kt^{0.5}$	Peppas-Sahlin ¹⁴⁰ $\frac{M_t}{M_\infty} = (k_1 t^n) + (k_2 t^{2n})$	Sigmoidal ¹¹⁹ $\frac{M_t}{M_\infty} = \frac{(A - B)}{1 + \exp\left(\frac{t - E}{C}\right)} + b$
Al-Zahrani ⁹² $\frac{M_t}{M_\infty} = 6(1 + \alpha) \left(\frac{tD}{\pi r^2}\right)^{\frac{1}{2}}$ $\alpha = \frac{C_\infty}{C_0 - C_\infty}$	Baker-Lonsdale ⁶⁸ $\frac{2}{3} \left[1 - \left(1 - \frac{M_t}{M_\infty}\right)^{\frac{2}{3}} \right] - \frac{M_t}{M_\infty} = k_H t$	Weibull ¹⁴⁹ $\frac{M_t}{M_\infty} = 1 - \exp(-kt^n)$
Korsmeyer-Peppas/Ritger-Peppas ¹³¹ $\frac{M_t}{M_\infty} = k_{kp} t^n$		

M_t : nutrient mass diffused up to time t ; M_∞ : nutrient mass diffused after infinite time (equilibrium); k , k_{kp} , k_0 , k_1 , and k : diffusion constants; k_H , k_2 : dissolution constants; a : initial nutrient released; b : release constant; n : diffusion exponent; D : diffusion coefficient, r : radius of the fertilizer granule; C_0 : initial concentration of nutrient; C_∞ : concentration of the fertilizer in the sphere at infinite time, A , B , C , and E : sigmoidal equation parameters.

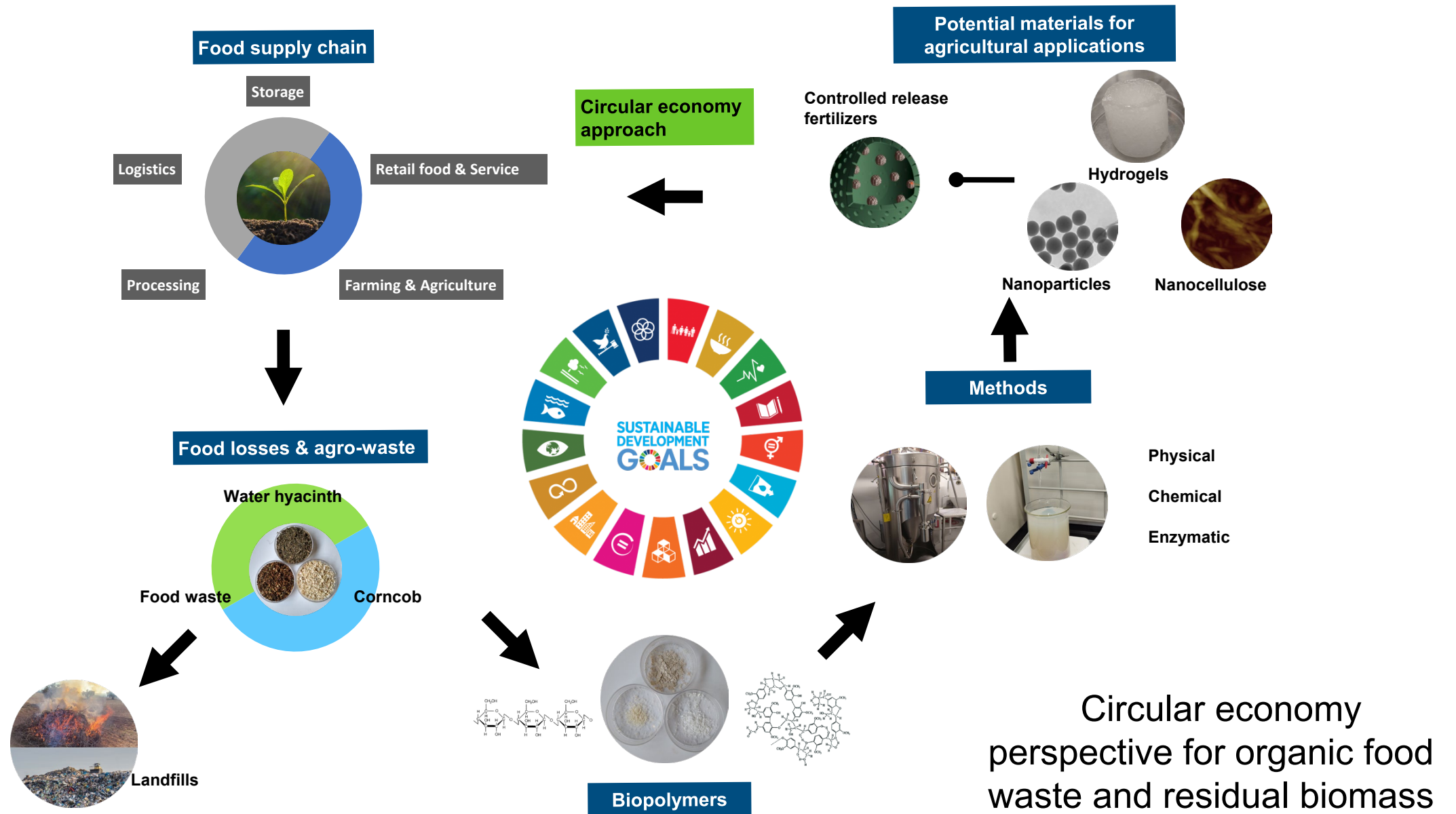
Absorption Routes for Nanoscale CRFs

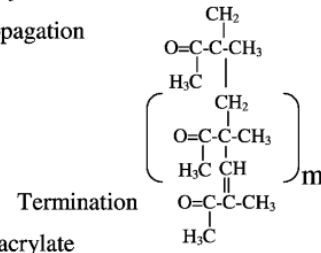
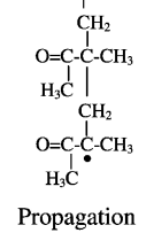
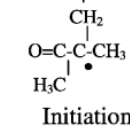
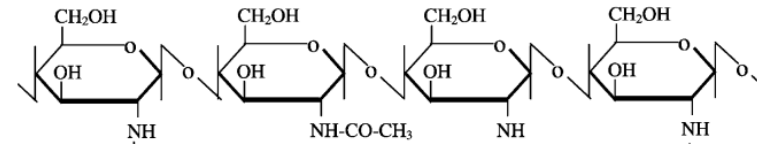
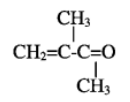
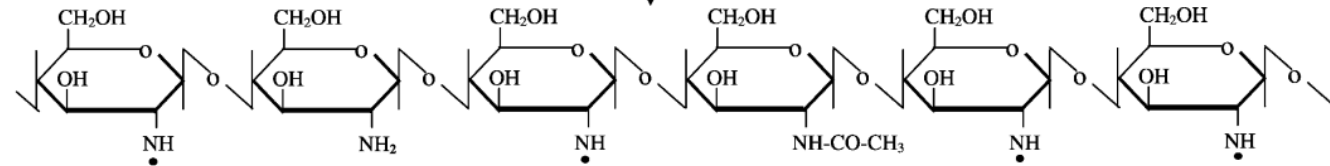
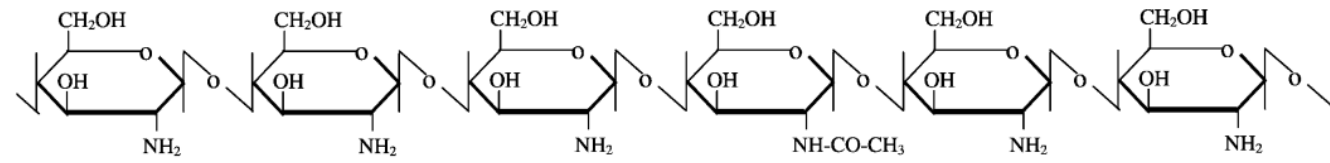


Potential Biowaste for Obtaining Controlled-Release Fertilizers (CRFs)



“Biodegradable materials come from a broad range of sources: agriculture (e.g., wheat straw, rice husk, starches, corn stover, corn cob, branches, sugarcane bagasse), forestry (e.g., forest litter, oat hull, birch wood), industrial activities (e.g., Kraft and sulfite liquors from pulp and papers), and food industry (e.g., leftovers, peels, waste frying oil, chicken residues). The value-added products that could be obtained from organic waste are dependent on the primary components present” (Gutierrez et al., 2022)





Chitosan-graft-polymethylmethacrylate

In the graft copolymer the presence of the carbonyl absorption peak at 1721 cm^{-1} confirmed the grafting reaction between chitosan and MMA

Chitosan is a biopolymer that reduces plant diseases through two main mechanisms: (1) Direct antimicrobial function against pathogens, including plasma membrane damage mechanisms, interactions with DNA and RNA (electrostatic interactions), metal chelating capacity, and deposition onto the microbial surface, (2) Induction of plant defense responses resulting from downstream signalling, transcription factor activation, gene transcription and finally cellular activation after recognition and binding of chitin and chitosan by cell surface receptors. This biopolymer have potential with capability to combating fungi, bacteria, and viruses phytopathogens.

