

Rheology and Self-Assembly of Carboxylated Cellulose Nanocrystals

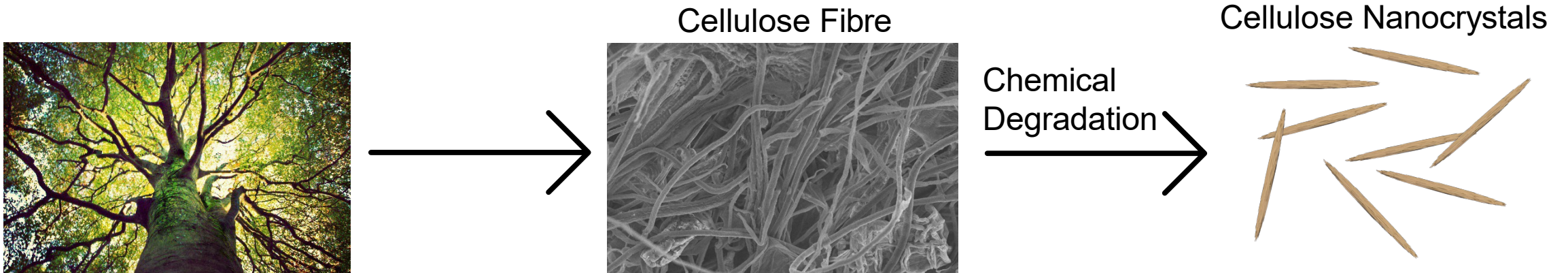
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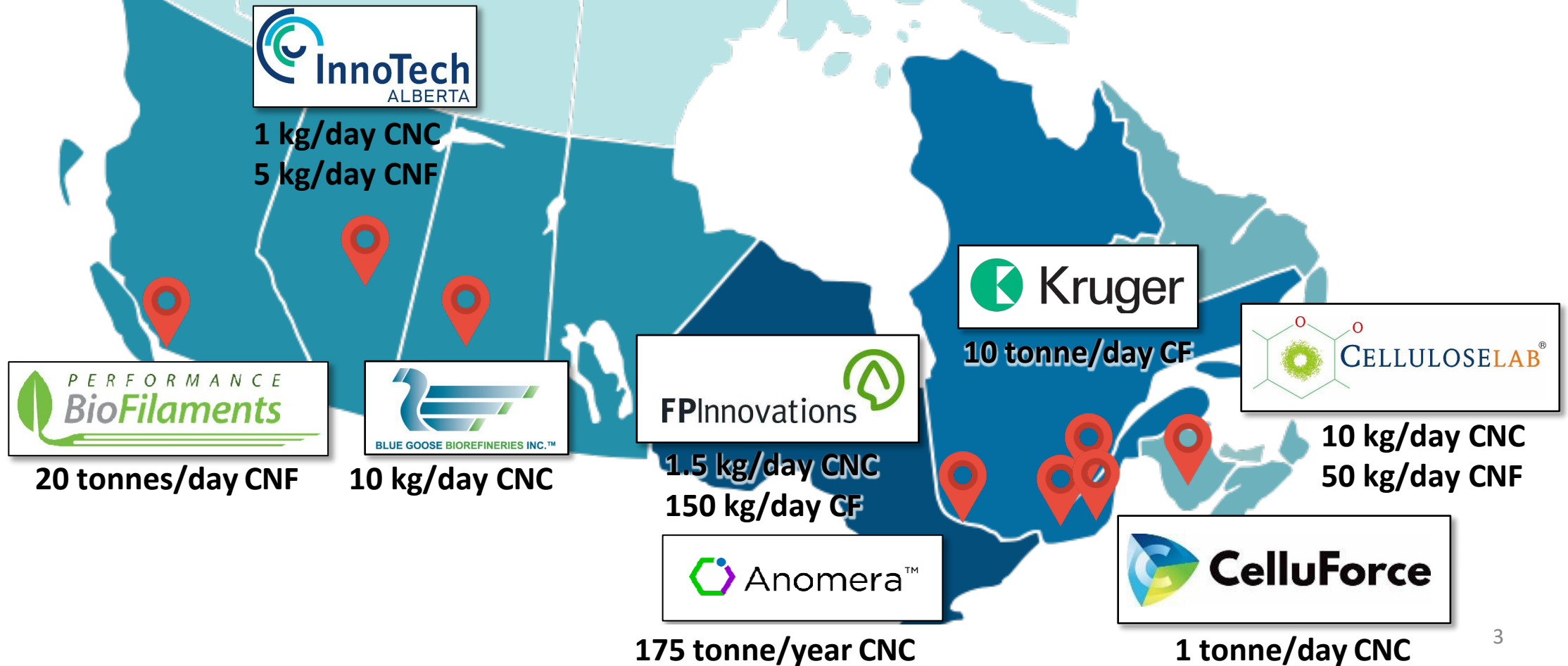


Cellulose Nanocrystals (CNCs)

- Cellulose nanocrystals are a renewable, bio-based nanomaterial with a high aspect ratio, high tensile strength, unique optical properties, and low density
- Rod-shaped particles chemically isolated from cellulose

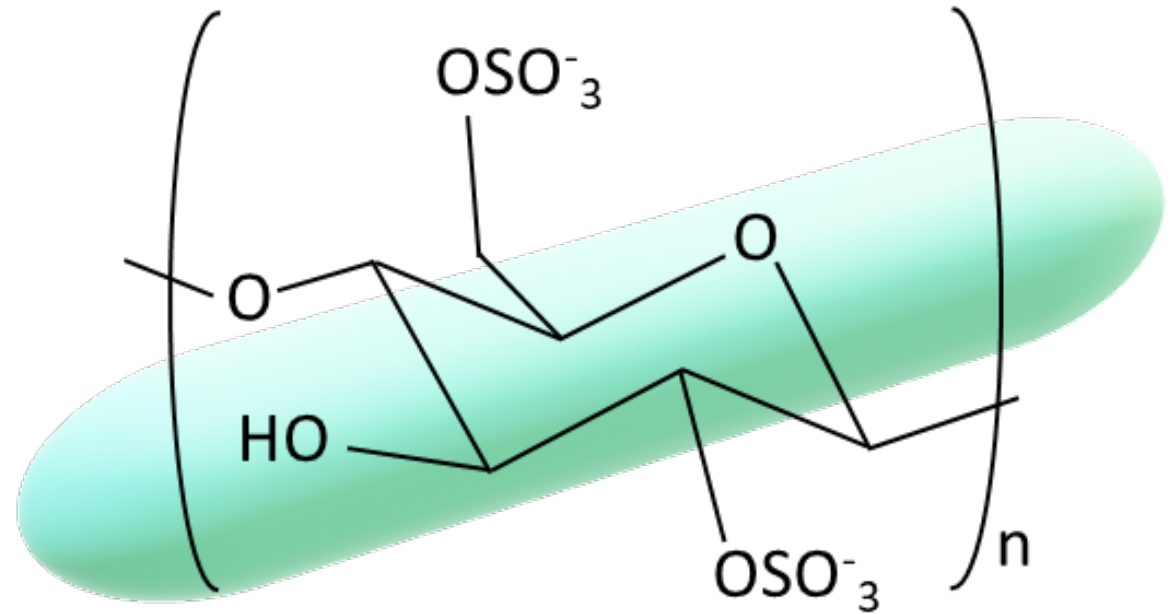


Canadian Nanocellulose Producers



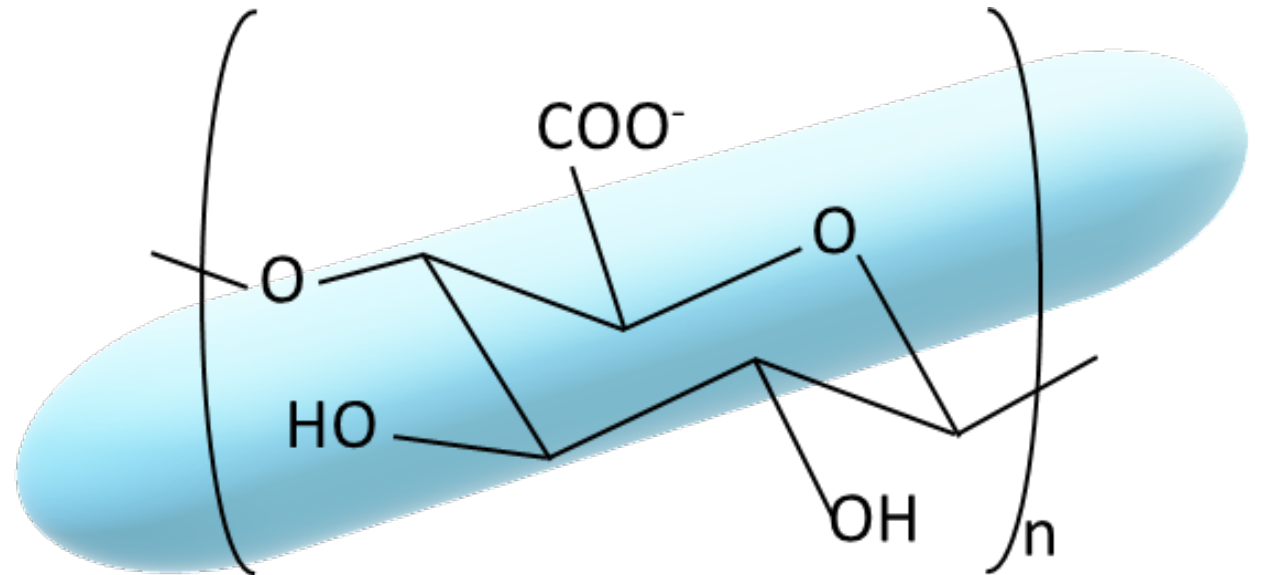
Sulfated CNCs

- CNCs are most often isolated through sulfuric acid hydrolysis which results in nanocrystals with charged sulfate half-ester surface groups
- Commercially produced
- Example applications:
 - Reinforcement additives in composite materials
 - Rheological modifiers in food, cosmetics and pharmaceuticals



Carboxylated CNCs

- Commercially produced by Anomera through oxidation using hydrogen peroxide
- Has carboxyl groups on the surface of the CNCs
- No existing work on the rheology of carboxylated CNCs
- Do they have the same useful properties as sulfated CNCs?
- Are they a potential alternative?

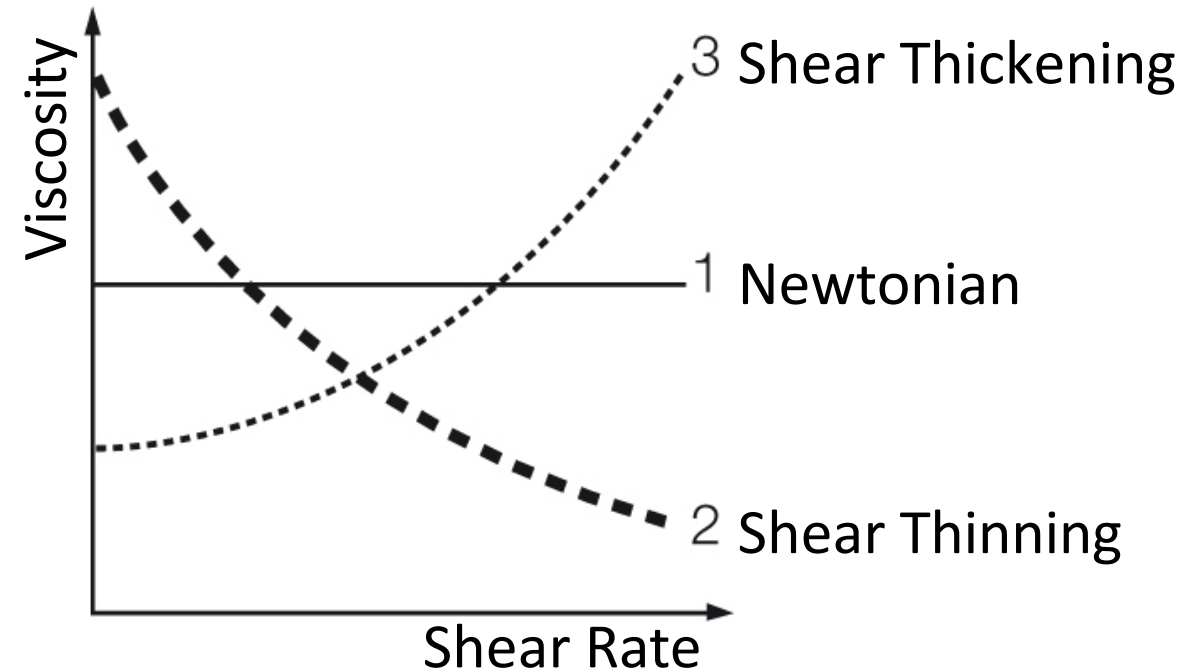


Rheological Applications

- Optimization of processing energy
- 3D printing – viscosity not too low not too high
- Lubrication – hydrodynamic lubrication
- Fast characterization of materials
- Useful rheological properties
 - Shear thinning behaviour – Rheological modifiers
 - Viscoelastic properties – lubrication

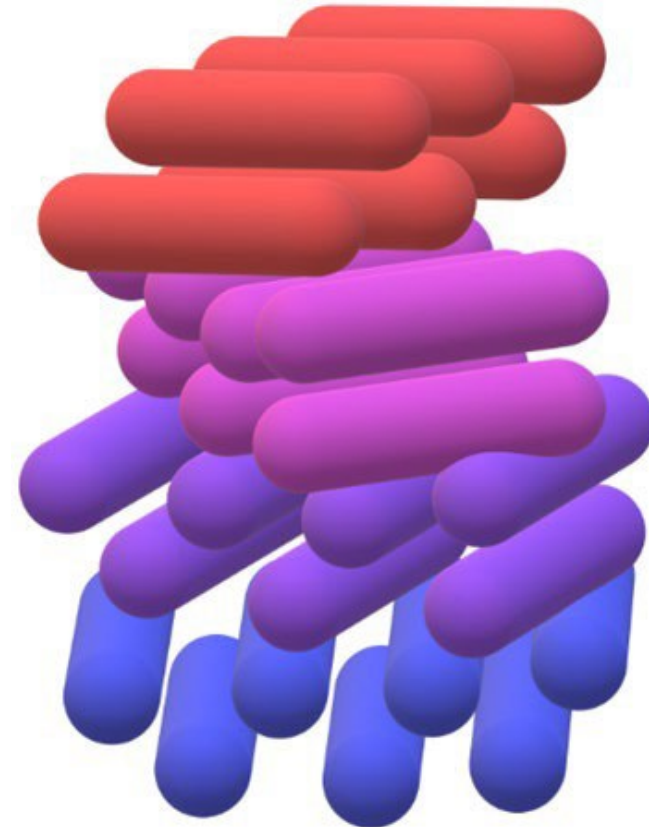


Types of material flow:



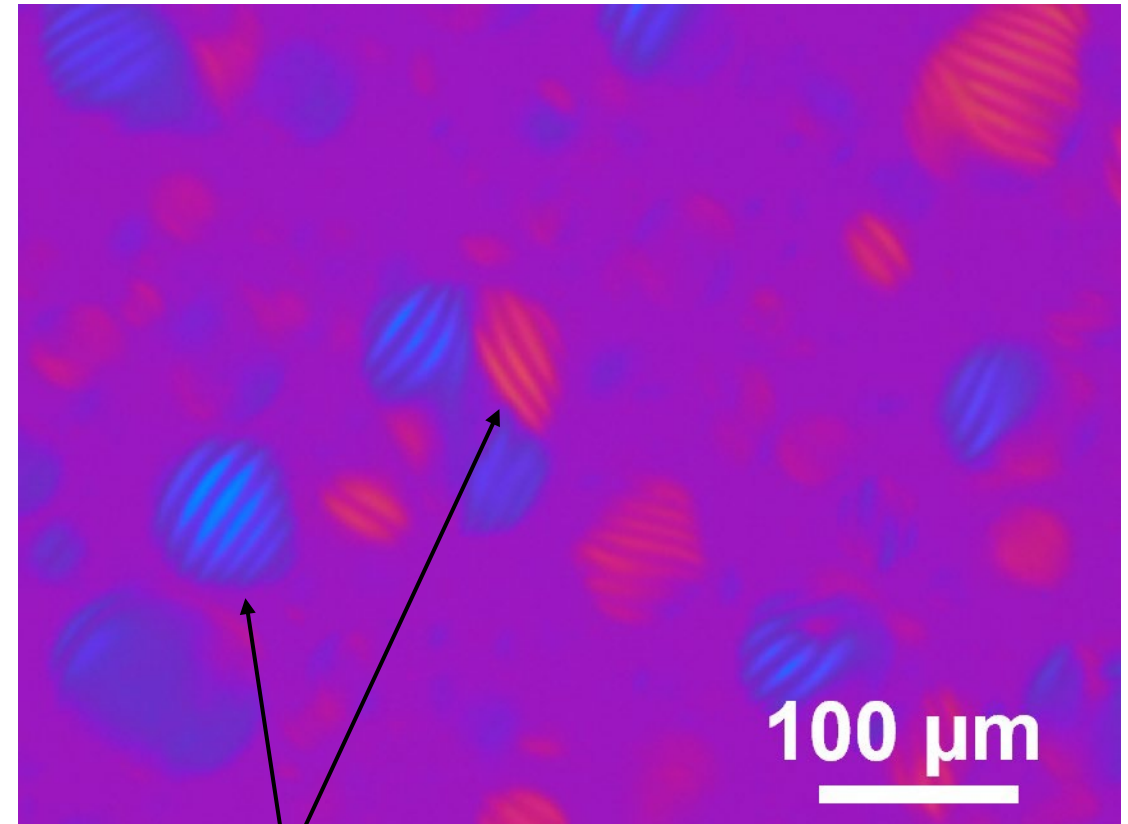
Self Assembly of CNCs

- The concentration above which self-assembly occurs
- CNCs arrange into chiral nematic liquid crystals
- Can be seen under polarized light



Self-Assembly Applications

- The self-assembly of sulfated CNCs are well defined^a
- Applications of CNC^b
 - Liquid-crystal phase optical filters
 - Optical films
 - Security paper
- cCNCs also self-assemble



Tactoids

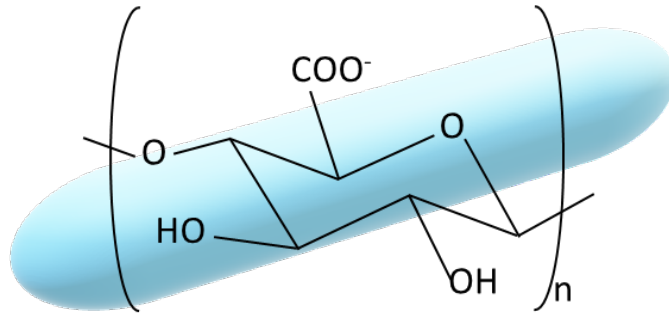
a. Lagerwall, Jan PF, et al. *NPG Asia Materials* 6.1 (2014) e80-e80.

b. <https://www.azooptics.com/Article.aspx?ArticleID=1406>

CNC Comparison

Property	Sulfated CNCs	Carboxylated CNCs
Critical concentration through phase separation		
Critical concentration through 3-region-flow		
Change of viscoelastic behaviour		

Characterization of Anomera's Carboxylated CNCs



	Low charge cCNCs	High charge cCNCs
Charge content (mmol COOH/ kg)	159 ± 3	217 ± 4
DLS apparent size (nm)	80 ± 1	81 ± 1

Methods of Determining Critical Concentration

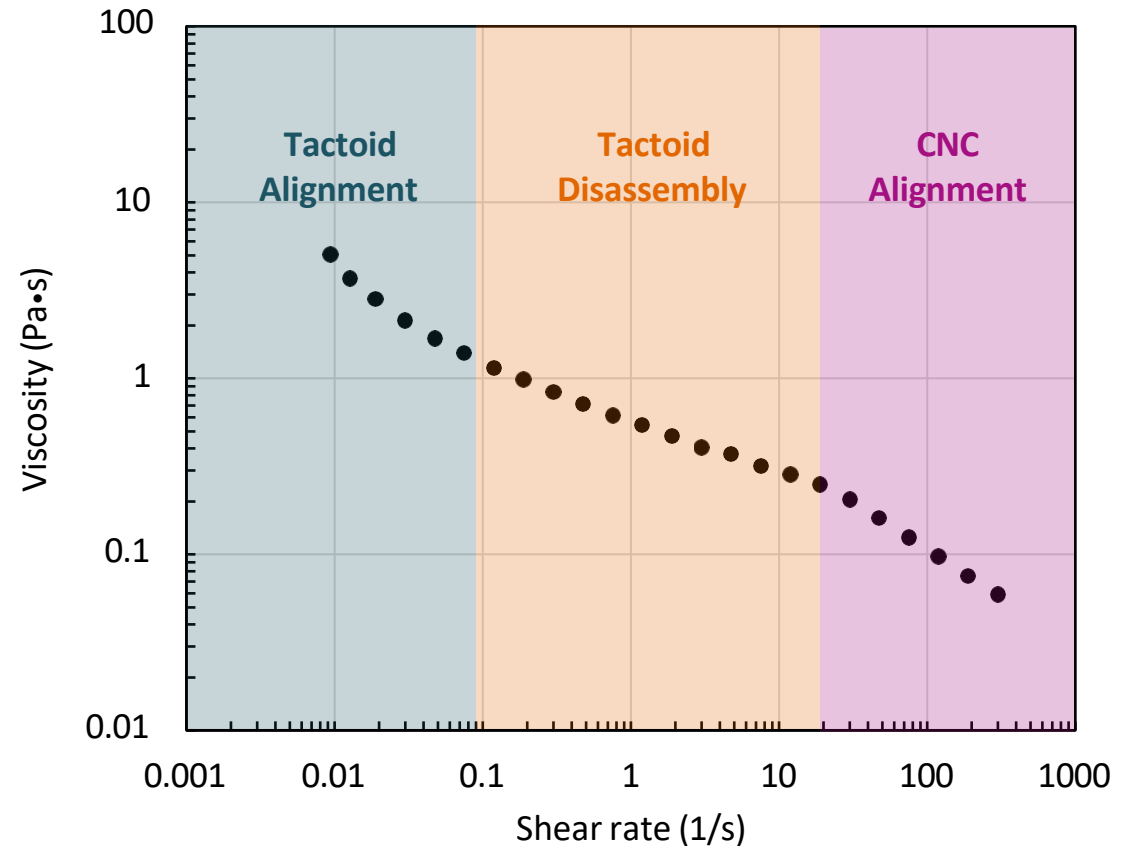
Phase Separation:

- Liquid crystalline region can be seen under polarized light



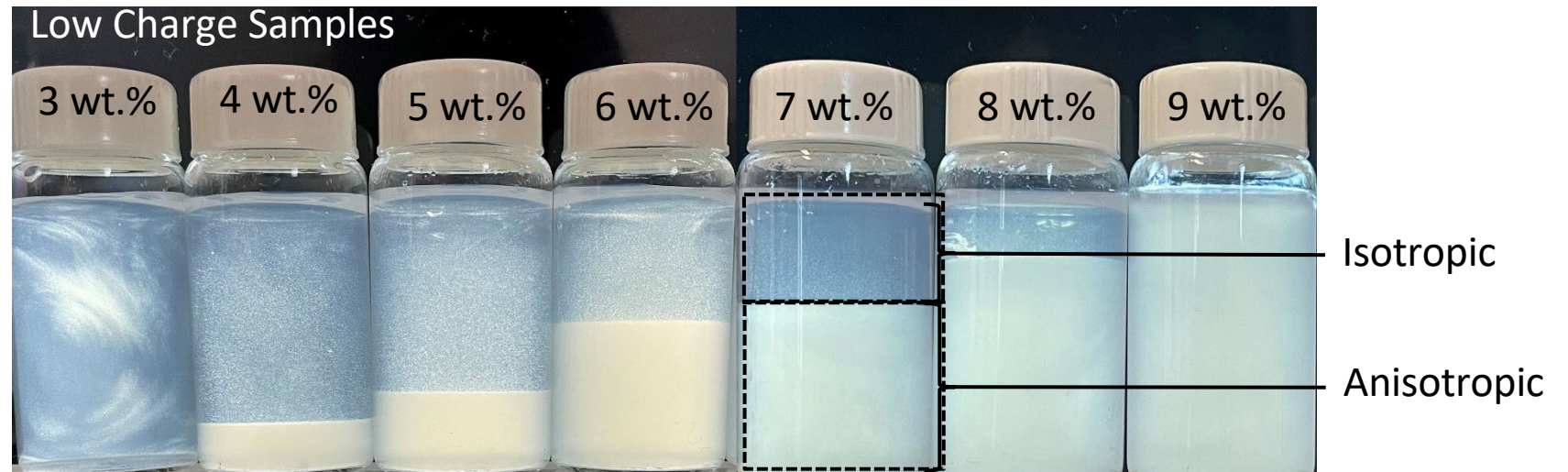
Rheology:

- The presence of liquid crystals results in three-region flow

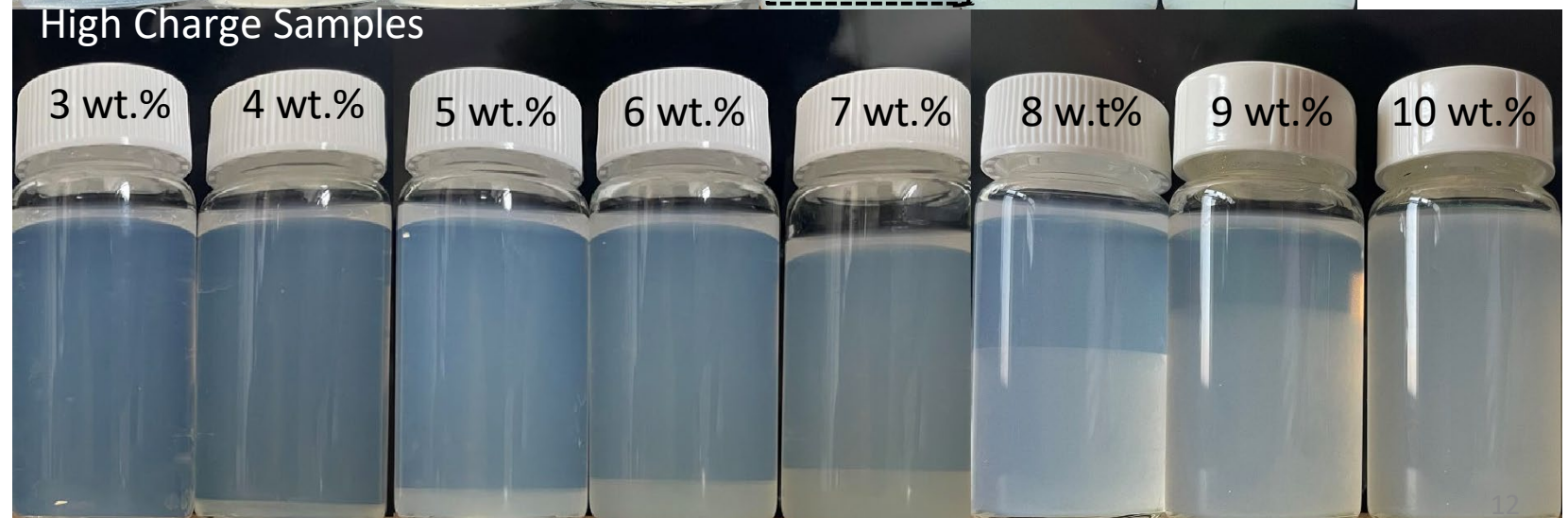


Critical Concentration Determination Through Phase Separation

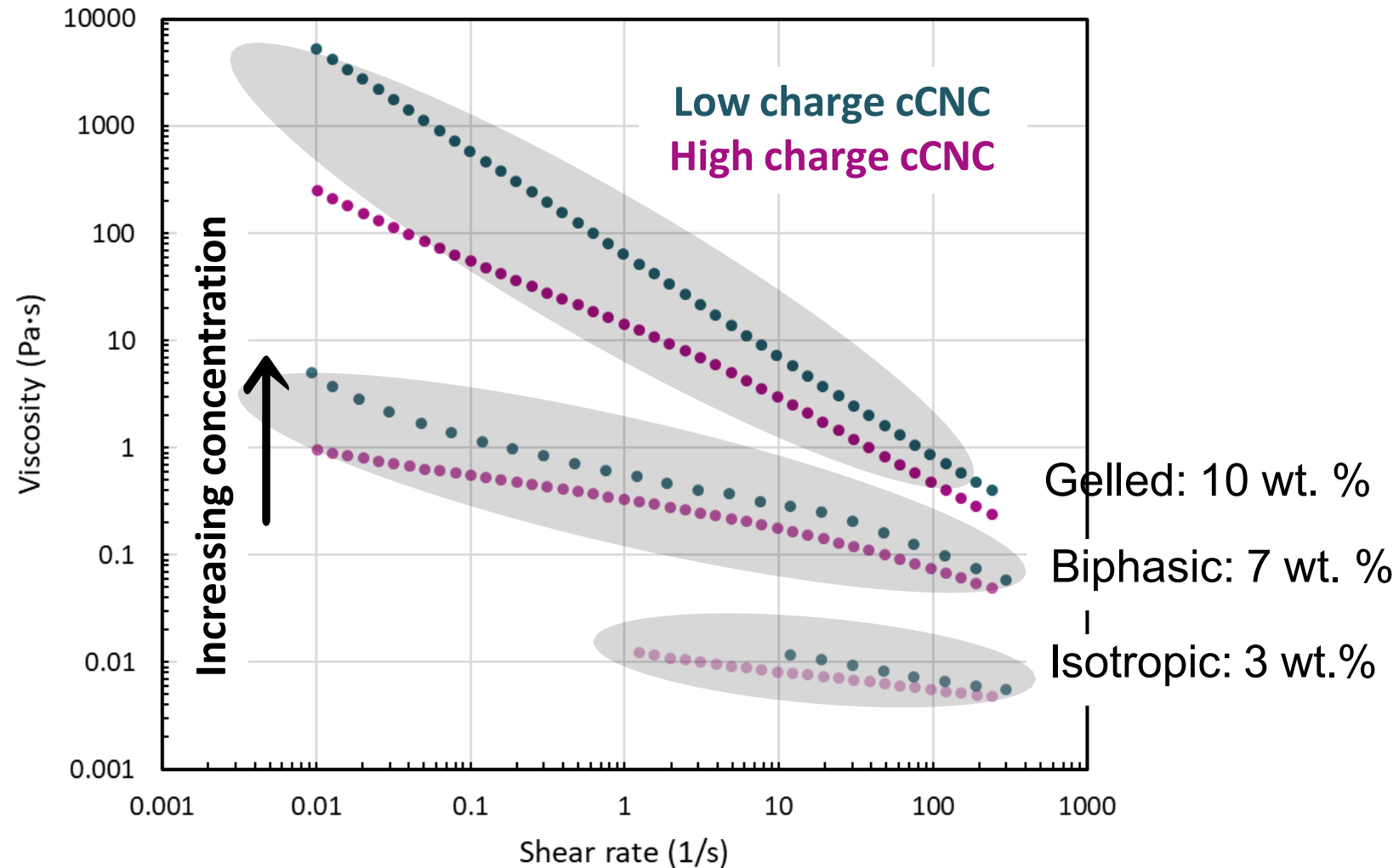
Critical concentration:
Low charge: 2.4 wt.%



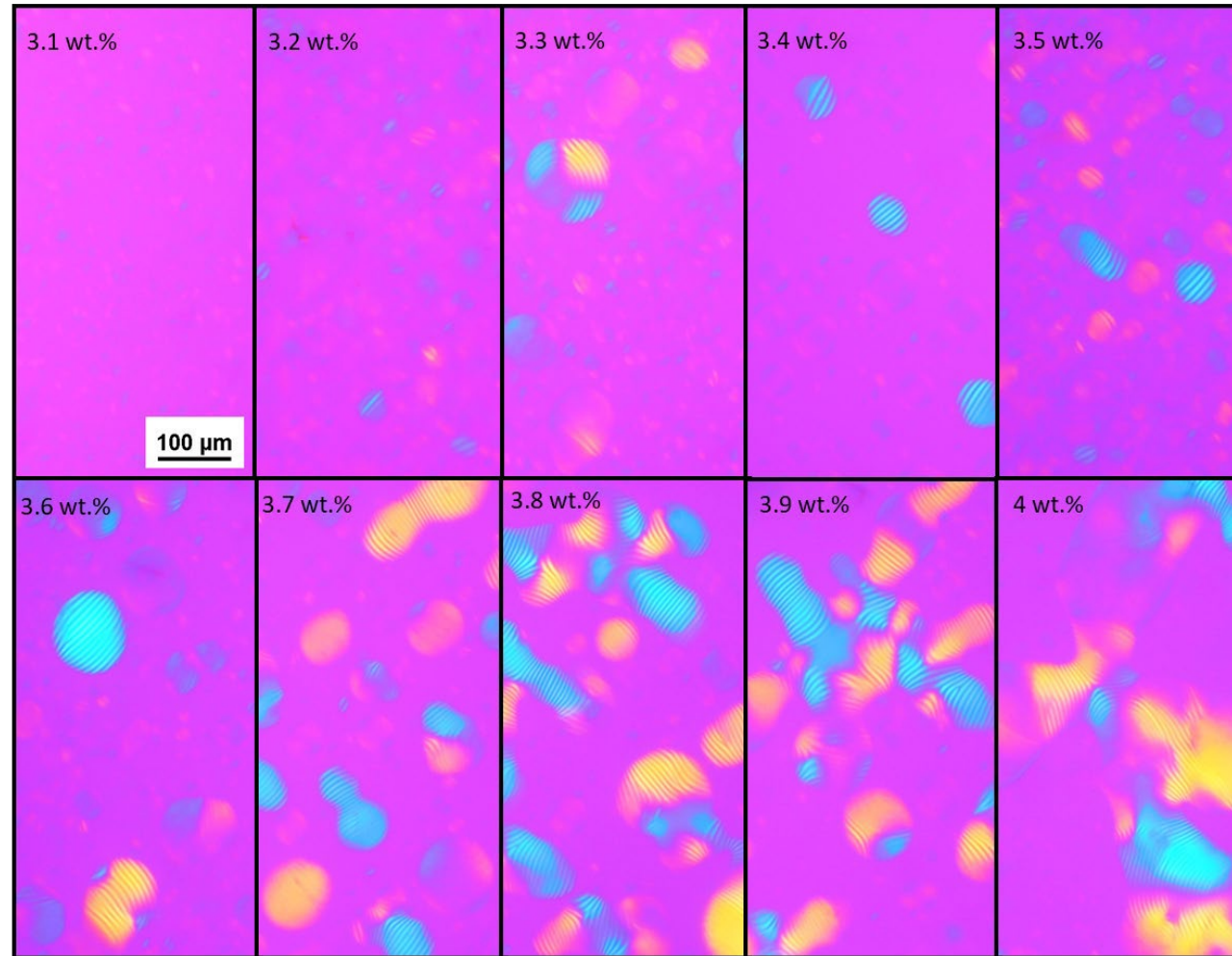
Critical concentration:
High charge: 3.6 wt.%



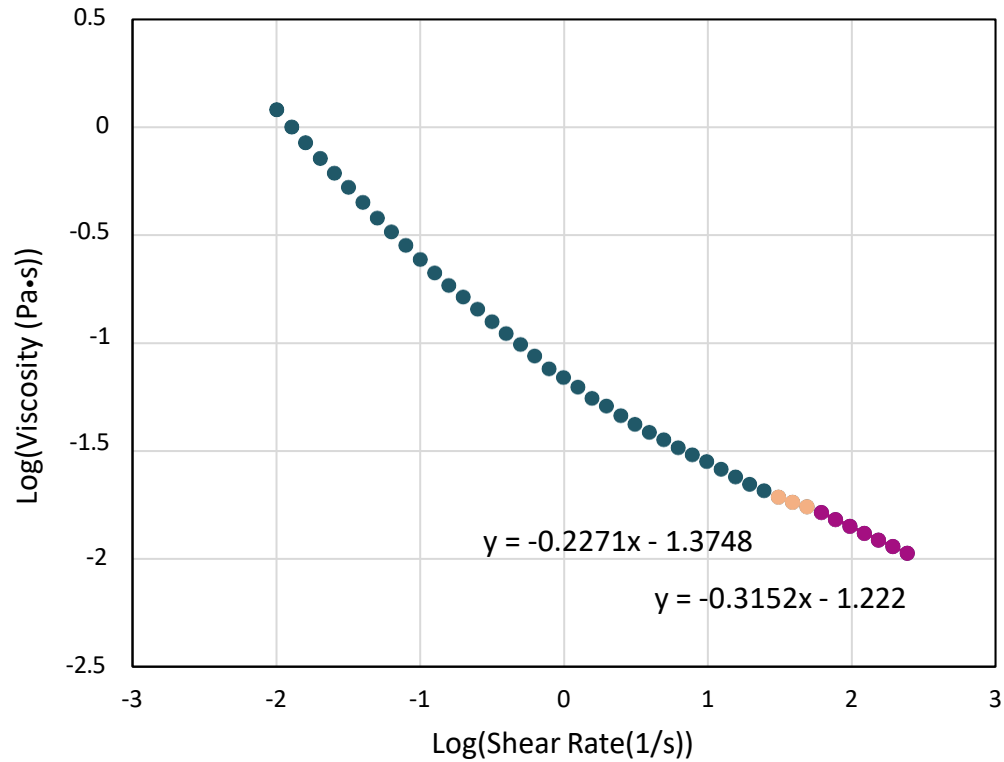
Critical Concentration Through Rheology



Critical Concentration Through Rheology – Tactoid Progression



Critical Concentration Through Rheology

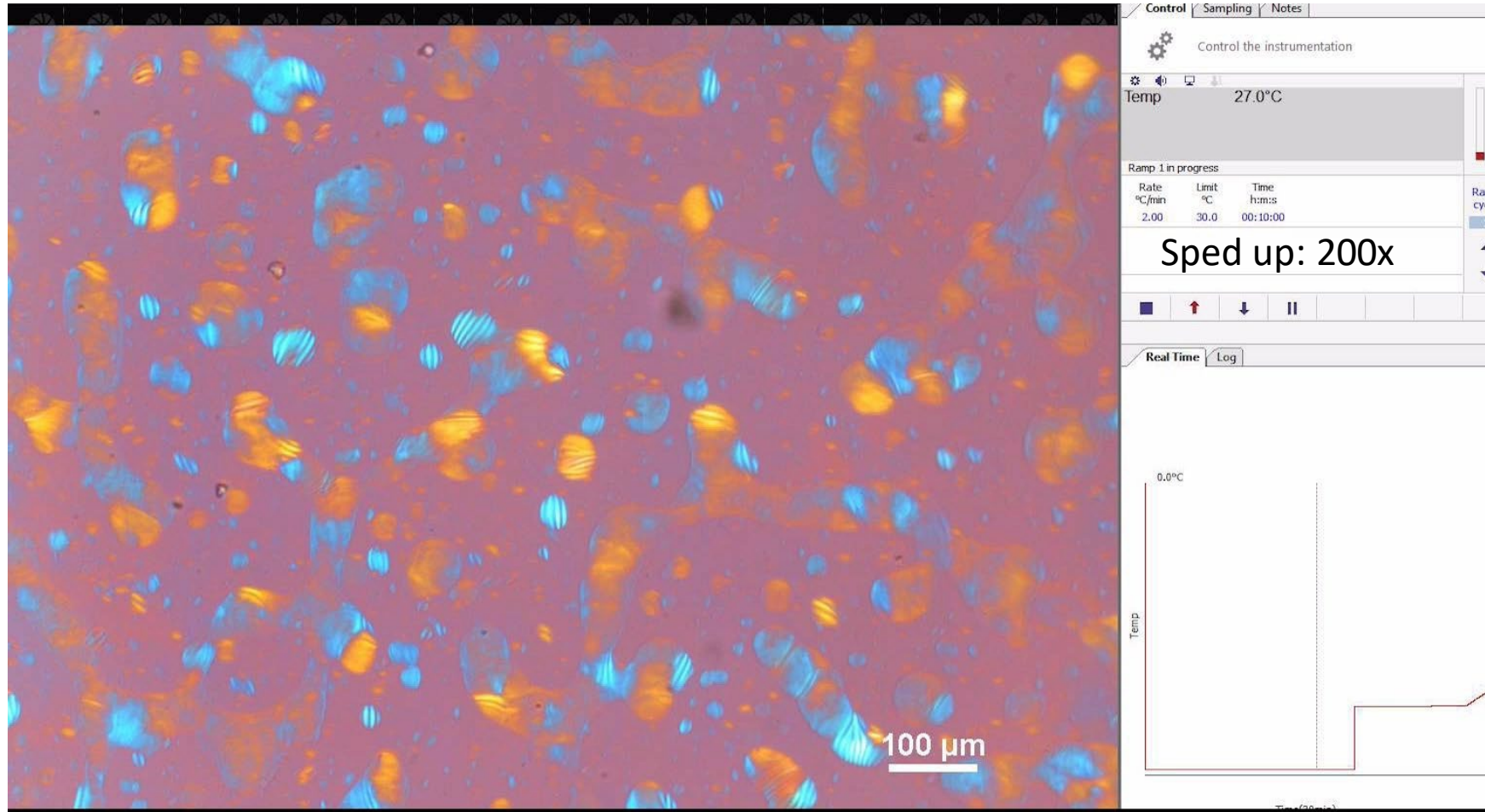


$$\sigma = K\dot{\gamma}^n$$

σ – Shear Stress
 $\dot{\gamma}$ – shear Rate

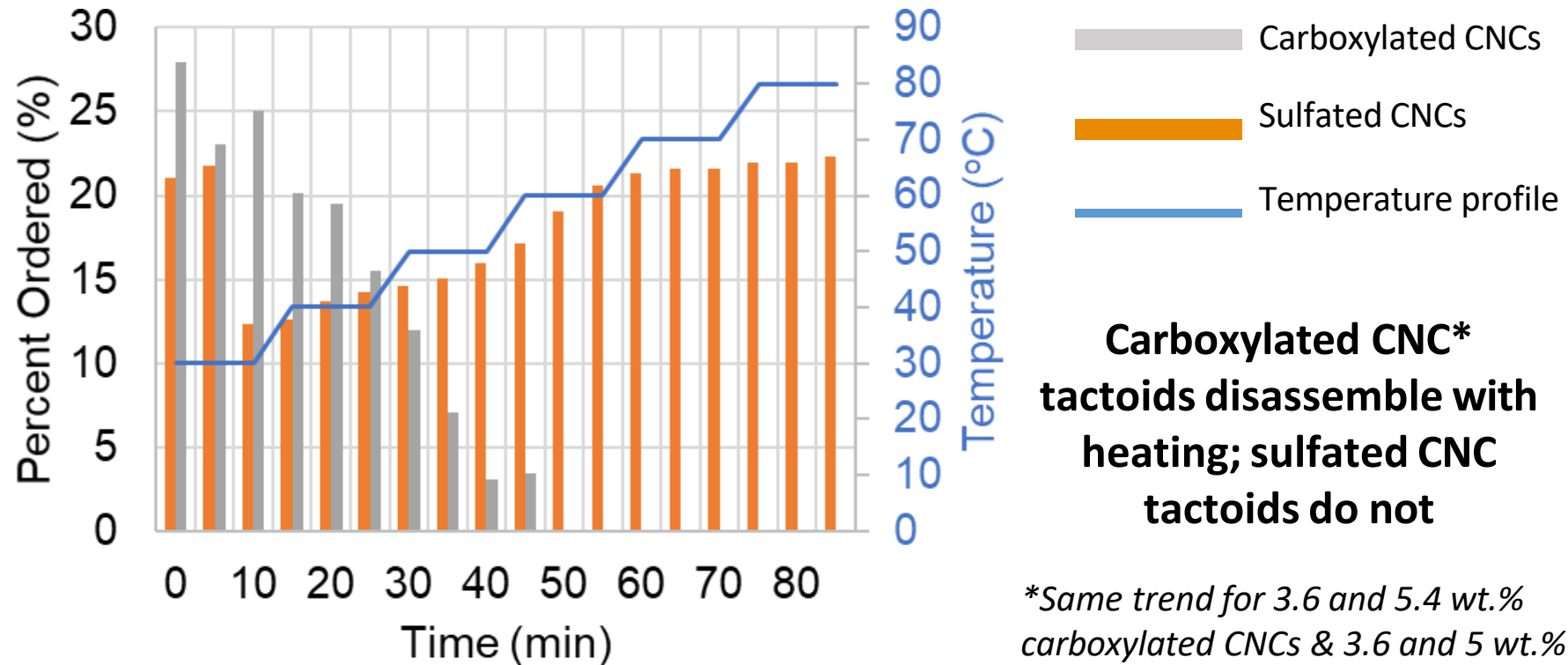
3.1 wt.%		3.2 wt.%		3.3 wt.%		3.4 wt.%		3.5 wt.%	
n_1	n_2	n_1	n_2	n_1	n_2	n_1	n_2	n_1	n_2
0.714	0.714	0.793	0.744	0.776	0.741	0.729	0.739	0.757	0.751
$\Delta n=0$		0.049		0.035		0.01		0.06	
3.6 wt.%		3.7 wt.%		3.8 wt.%		3.9 wt.%		4.0 wt.%	
n_1	n_2	n_1	n_2	n_1	n_2	n_1	n_2	n_1	n_2
0.766	0.739	0.764	0.740	0.756	0.736	0.737	0.690	0.773	0.685
$\Delta n=0.027$		0.024		0.02		0.047		0.088	

Tactoid Disassembly With Heating



- Some tactoids disappear quickly
- Some merge
- Others go out of focus
- By 50 °C there are no tactoids left

Tactoid Disassembly with Heating



**Same trend for 3.6 and 5.4 wt.% carboxylated CNCs & 3.6 and 5 wt.% sulfated CNCs*

Critical Concentration Summary – Sulfated and Carboxylated CNCs

Property	Sulfated CNC	Carboxylated CNC
Critical concentration through phase separation	3 wt.% ^a 0.6 wt.% ^b 4.3 wt.% ^c	2.4 wt.% (low charge) 3.6 wt.% (high charge)
Critical concentration through 3-region-flow	0.6 wt.% ^b 4.8 wt.% ^c 0.4 wt.% ^d 1.5 wt.% ^d 3 wt.% ^e 4 wt.% ^e	4 wt.% (low charge) 5 wt.% (high charge)
Change of viscoelastic behaviour	Next	

a. Honorato-Rios, Camila, et al. Frontiers in Materials (2016): 21.

b. Qiao, Congde, et al. Food Hydrocolloids 55 (2016): 19-25.

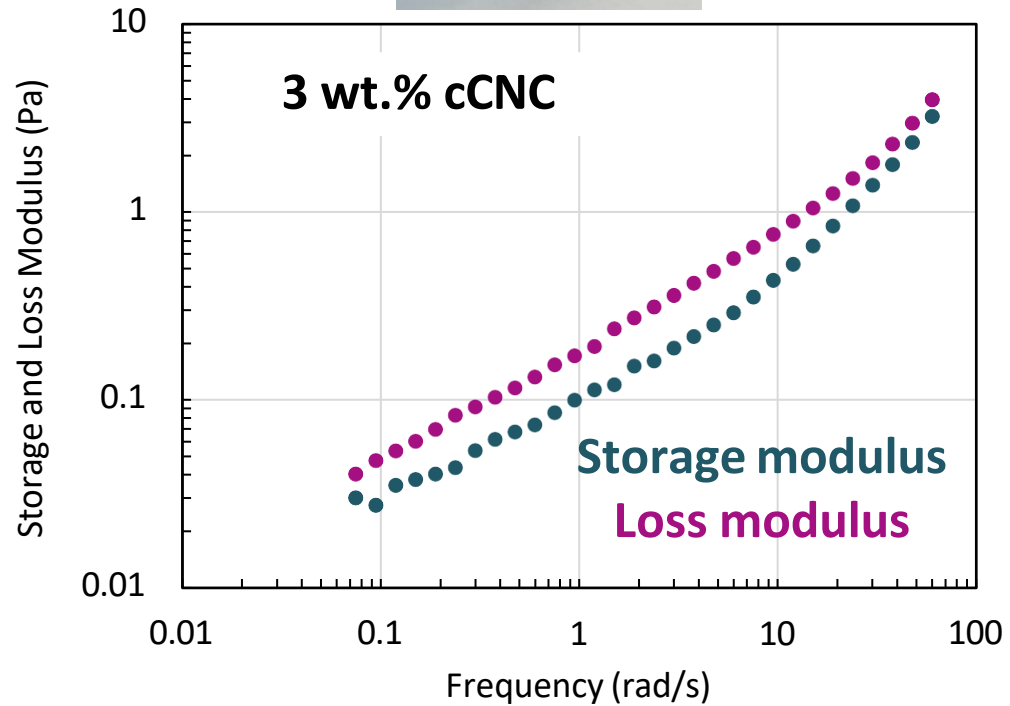
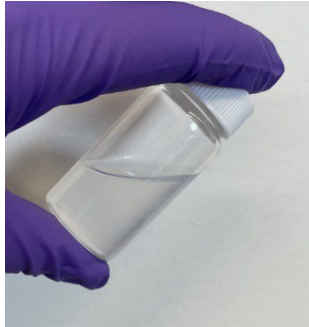
c. Liao, Jianshan, et al. Cellulose 27.7 (2020): 3741-3757.

d. Wu, Qiang, et al. Journal of Applied Polymer Science 131.15 (2014).

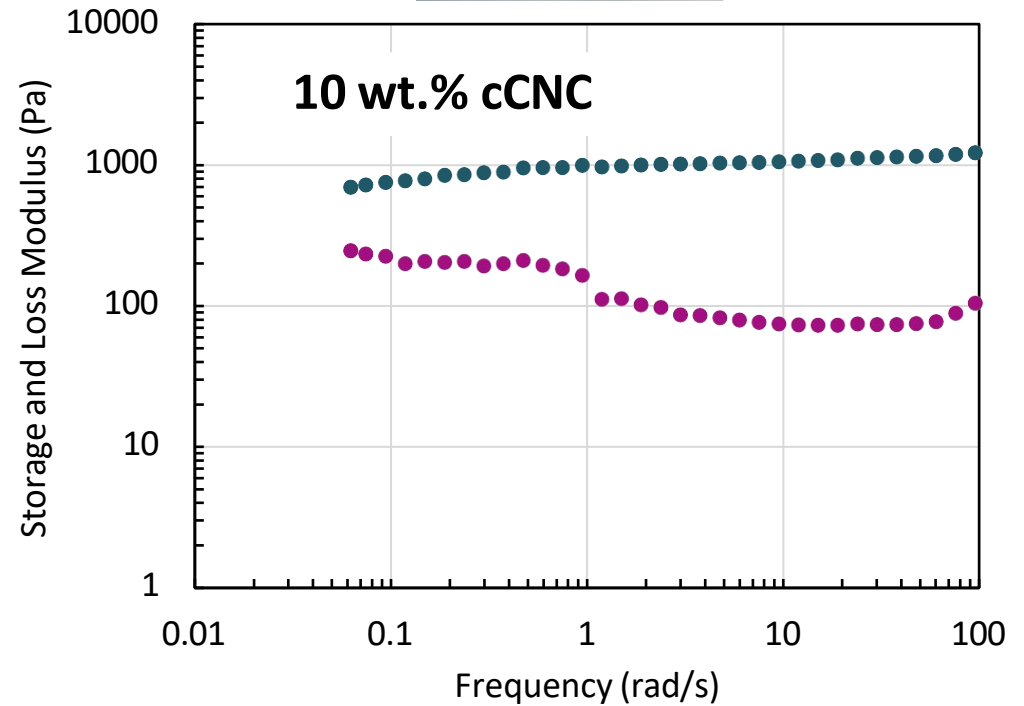
e. Shafeiei-Sabet, Sadaf, et al. Rheologica Acta 52 (2013): 741-751.

Storage and Loss Modulus of cCNCs

Storage modulus < Loss modulus



Storage modulus > Loss modulus



Summary of Viscoelastic Properties of Sulfated and Carboxylated CNCs

Property	Sulfated CNC	Carboxylated CNC
Critical concentration through phase separation	0.6 - 4.3 wt.% ^{a,b,c}	2.4 wt.% (low charge) 3.6 wt.% (high charge)
Critical concentration through 3-region-flow	0.6 - 4 wt.% ^{b-e}	4 wt.% (low charge) 5 wt.% (high charge)
Change of viscoelastic behaviour	3 wt.% ^b	7 wt.% (low charge) 8 wt.% (high charge)

a. Honorato-Rios, Camila, et al. *Frontiers in Materials* (2016): 21.

b. Qiao, Congde, et al. *Food Hydrocolloids* 55 (2016): 19-25.

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Summary

Property	Sulfated CNC	Carboxylated CNC
Critical concentration through phase separation	0.6 - 4.3 wt.% ^{a,b,c}	2.4 wt.% (low charge) 3.6 wt.% (high charge)
Critical concentration through 3-region-flow	0.6 - 4 wt.% ^{b-e}	4 wt.% (low charge) 5 wt.% (high charge)
Change of viscoelastic behaviour	3 wt.% ^b	7 wt.% (low charge) 8 wt.% (high charge)

- Carboxylated CNCs have a viscosity in the same range as sulfated
- Shear thinning behaviour is present
- The critical concentrations are in the same range as sulfated CNCs
- Viscoelastic properties are not in the same range

Carboxylated CNCs shows two critical concentrations: phase separation and impact on flow

a. Honorato-Rios, Camila, et al. *Frontiers in Materials* (2016): 21.
 b. Qiao, Congde, et al. *Food Hydrocolloids* 55 (2016): 19-25.
 c. Liao, Jianshan, et al. *Cellulose* 27.7 (2020): 3741-3757.
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 e. Shafeiei-Sabet, Sadaf, et al. *Rheologica Acta* 52 (2013): 741-751.

International Conference on Nanotechnology for Renewable Materials



Thank You

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